



Evaluation of Secondary Metabolites of *Ageratina adenophora* and Synthesis of Silver Nanoparticles for its Antibacterial and Antioxidant activity

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ABSTRACT

Synthetic antibiotics have been successfully utilized for decades against pathogenic bacteria to control infectious diseases. However, the continuous and overuse has resulted in multidrug resistant (MDR) bacterial species. Further, the negative side effects caused by commercial antibiotics also hindered their usage. The phytochemicals produced by plants in response to adverse biotic and abiotic conditions possess significant pharmacological properties and can be an effective alternative to synthetic antibiotics. The phytochemicals of *Ageratina adenophora*, served the role of reducing and stabilizing agent. *Ageratina adenophora* mediated silver nanoparticles (Aa-AgNPs) were characterized using advanced spectroscopic instrumentation. The qualitative analysis by GC-MS showed Methyl ionone, 2(3H)-Naphthalenone, 4, 4a, 5, 6, 7, 8- hexahydro-4a, 7, 7-trimethyl-(R), Isolongifolone as the major compounds. The quantitative estimation showed leaves were rich in total phenol, flavonoids, alkaloids and tannins. The Aa-AgNPs were effective in inhibiting bacterial pathogens. Further, *A. adenophora* mediated nanoparticles possessed strong antioxidant activity.

Keywords: *Ageratina adenophora*, Antibacterial agent, Antioxidants, Isolongifolone, Silver nanoparticles.

INTRODUCTION

Bacterial infections pose a significant threat to humans around the world^{1,2}. The factors that contribute to microbial infections in emerging and underdeveloped countries include lack of public awareness about public hygiene, environmental

cleanliness, effective sanitation, lack of critical primary health care, and rapid population growth rate³. *Escherichia coli*, *Streptococcus pneumoniae*, *Staphylococcus aureus*, *Pseudomonas aeruginosa*, and *Klebsiella pneumoniae* are the important pathogenic bacterial species that cause frequent infection in humans. These microorganisms cause



childhood infections and premature deaths⁴. Synthetic antibiotics are effective against bacterial pathogens. However, the emergence of bacterial resistance to commercial antibiotics hinders the efficacy of treatment. Furthermore, negative side effects of antibiotics such as allergies, hypersensitivity, and immunosuppression also play critical role in treating infectious disorders. The overuse and improper application of antibiotics against microbes led to the emergence of multidrug-resistant microorganisms⁵⁻⁷. This crisis would worsen in the near future if alternative strategies that are secure are not developed.

Medicinal plants are preferred, as they produce phytochemicals (secondary metabolites) that possess effective pharmacological properties. The common people still prefer medicinal plants as they are safe nature and lack of side effects, despite technological advancements in contemporary antibiotic drugs and treatment approaches. Phytochemicals such as alkaloids, flavonoids, terpenoids, saponins, phenolic compounds, and sterols efficiently attenuate the side effects and can be a safe alternative to synthetic antibiotics⁸. Previous research has demonstrated the efficacy of phytochemicals against antibiotic-resistant bacterial pathogens⁸⁻¹². Recent technological advances have paved the way for the identification of diverse range of metabolites that play an effective role in the survival of plants under extreme biotic and abiotic environments¹³. Since more than 60,000 years ago, traditional medicine in India uses phytochemicals^{14,15}. In comparison to conventional antimicrobials, plant-derived bioactive chemicals have multiple mechanisms of action against microbial infections, including binding affinity for numerous targets.

Reactive oxygen species (ROS) is required for normal cellular activity. But when the intracellular concentration of ROS rises, it causes a slew of negative consequences, including cancer, diabetes, inflammation, premature ageing, and atherosclerosis. Antioxidants are molecules that protect the body from the negative effects of reactive oxygen species (ROS). Secondary metabolites produced including flavonoids, anthocyanins, and carotenoids, are effective free radical scavengers. Among phytochemicals found in plants, phenolic compounds are antioxidative agents¹⁶. They breakdown peroxides, donate hydrogen, quench singlet and triplet oxygen and trap free radicals.

A. adenophora (Asteraceae) is an invasive plant. The plant originated in Mexico, although it can also be found in nations in South and Southeast Asia. Larvicidal¹⁷, antimicrobial¹⁸, anti-inflammatory¹⁹, antipyretic²⁰, wound-healing²¹, antioxidant²², and analgesic²³ have been demonstrated. *E. coli*, *Bacillus subtilis*, *S. aureus*, *K. pneumoniae*²⁴, and *Proteus mirabilis* were effectively inhibited by *A. adenophora*²⁵.

Nanotechnology, dynamic field of science involves biology, physics and chemistry of nanoscale materials that has unplugged new avenues in the diagnosis and treatment of diseases²⁶. Silver is among the most effective metallic nanoparticles used in medicine. Silver nanoparticles (AgNPs) exhibit biological activity, including antibacterial, antioxidant, antifungal, DNA fragmentation²⁷ and cytotoxicity²⁸ properties. This study focuses on fabricating AgNPs from *A. adenophora aqueous* extract of leaves as an alternative strategy for enhancing the biological activity of phytochemicals in the control of microbial pathogens and in quenching free radicals.

MATERIALS AND METHODS

Plant material

Fresh healthy *A. adenophora* leaves collected from Kodaikanal hills, Tamilnadu, India. The washed leaves were shade dried for 10 days and powdered using mechanical blender.

Preparation of solvent extracts

About 20 g of powdered plant material was packed into a thimble and extracted with 250 mL of different solvents (water, petroleum ether, benzene, chloroform, acetone and methanol) in a soxhlet's apparatus. After 24 h of extraction the solvent was evaporated under reduced pressure. Crude extract was collected and stored at 4°C²⁶.

Determination of ash values

The physicochemical parameters such as percentage of ash and extractive values and weight loss on drying were carried out using the official methods^{29,30}.

Water soluble extracts

A 5% aqueous plant extract was prepared and filtered using Whatmann filter paper. An empty evaporating dish was weighed, and 25 mL of a 5%

plant extract was added and heated until a damp mass was formed. The damp mass was cooled and weighed. The weight of empty dish to that of damp mass gives the value of water-soluble extract.

Alcohol soluble extracts

A 5% methanol plant extract was prepared and filtered using Whatmann filter paper. An empty evaporating dish was weighed, and 25 mL of a 5% plant extract was added and heated until a damp mass was formed. The damp mass was cooled and weighed. The difference in weight of empty dish and with damp mass gives the value of alcohol-soluble extract.

Phytochemical analysis

The initial phytochemical screening assays were conducted with the established techniques of Kokate³¹ and Harborne³². Approximately 100 g of leaf powder was extracted using the Soxhlet apparatus for 24 h with water, benzene, petroleum ether, acetone, chloroform, and methanol. The filtrate was concentrated and diluted in respective solvents. Solvent extracts were analysed for phytochemicals.

GCMS analysis for identification of compounds

Plant powder and methanol (1:25 ratio) was subjected to GC-MS analysis for the identification of compounds using the procedure adapted by³³. 1 µl of the sample was loaded into fused silica column coated with poly dimethyl siloxane (stationary phase) and the mobile phase, helium was applied (1 mL/minute). Injector temperature was set to 250°C and the oven temperature was increased to 250°C from its initial temperature of 60°C at the rate of 2°C/min and maintained for 5 minute. The ionization voltage used was 70 eV with a split ratio of 1:25. Retention time/mass spectra of known compounds and published data were utilised to identify compounds.

Determination of total phenol content

Methanol extract of *A. adenophora* leaves was used to quantify phytochemicals. The total phenol³⁴ and tannins³⁵ were determined by Folin-ciocalteau method. Total flavonoid concentration was determined using a modified calorimetric method³⁶. Following the protocol of³⁷ the alkaloid content was determined.

Ageratina adenophora mediated silver nanoparticles (Aa-AgNPs)

Slight adjustments were made to the

methodology of Gautam *et al.*,³⁸ in the preparation of Aa-AgNPs. Leaf powder (5 g) and distilled water (100 mL) was heated for 30 min at 70°C, cooled and filtered (Whatman filter paper. No. 1). For the reduction reaction, 50 mL of aqueous leaf extract was added to 50 mL of 3 mM AgNO₃ solution and stirred vigorously with a magnetic stirrer for 30 minutes. The variation in colour of the reacting solution indicated Aa-AgNPs synthesis. The resulting solution was then centrifuged at 15,000 revolutions per minute for 10 minutes. The pellets obtained were dispersed in deionized water to eliminate uncoordinated biological molecules. The pellets were then dried in a lyophilizer.

Fourier transform infra-red spectroscopy

FTIR spectroscopy was used in identifying the functional phytochemical groups. The nano solution was centrifuged at 60,000rpm for 40 min, after which the pellets were dissolved in deionized water and filtered through 0.45 m Millipore filter paper. 1 mg of Aa-AgNPs and 10 mg of KBr pellets were, finely crushed, and formed into a pellet using hydraulic pressure. The spectra were captured in the region of 4000 to 400 cm⁻¹.

X-ray diffraction analysis

XRD analysis confirms the synthesis of Aa-AgNPs (PAN analytical X pert PRO Model) with operating conditions of voltage of 40 kV; current of 30 mA with CuK radiation. Particles size (L) was calculated with Debye-Scherrer's equation.

$$L = 0.9\lambda / \beta h\theta$$

λ = wavelength of the X-ray, β = full width and half maximum and θ = the Bragg's angle.

Scanning Electron Microscopic (SEM) analysis

The morphology and distribution of Aa-AgNPs were investigated with a FESEM, Carl Zeiss-Sigma model, Germany. Aa-AgNPs were placed on a double-sided metal stub and sputtered with gold (Au) for SEM investigation.

Antibacterial activity of Aa-AgNPs

E. coli, *K. pneumoniae* (MTCC1610), *S. aureus* (MTCC96) and *S. pneumoniae* (MTCC1610) were procured from IMTECH, India. Antibacterial activity was determined using agar well diffusion assay³⁹. The nutrient broth medium was

seeded with selected bacteria, and cultures were adjusted to 0.5 McFarland standards (1 10⁸ CFU mL⁻¹) and plated on sterile nutrient agar. The plates were dried for 15 min before to sensitivity testing. Aa-AgNPs (40, 50, and 60 µg/mL) and a standard antibiotic (20 µL) were evaluated for their efficacy. Amoxicillin was employed as a positive control. The plates were then incubated at 37°C for 24 h and bacteriostatic zone was measured.

***In-vitro* antioxidant assay**

DPPH (2,2-diphenyl-1-picrylhydrazyl), ABTS (2, 2'-azino-bis (3-ethylbenzothiazole-6-sulphonic acid), nitric oxide scavenging assays, and total reducing power assays were utilised to test the antioxidant property of the synthesised nanoparticles. The DPPH and total reducing power analysis were conducted according to Syed Ismail *et al.*,⁴⁰. The nitric oxide was investigated following Hashemi and Ebrahimzadeh⁴¹. The inhibition of ABTS by Aa-AgNPs was determined using Hajebi *et al.*,⁴² technique.

The inhibition(%) or scavenging of free radicals was calculated by the formula:

$$\% \text{inhibition} = (A_0 - A_1) / A_0 \times 100$$

A₀=control and A₁=sample

RESULT AND DISCUSSION

Percentage of yield obtained with solvent extracts was presented in Table (1). The aqueous extract had a high yield percentage of 4.105% and low yield was recorded with benzene (1.215%). The yield by different solvent extract was as follows: aqueous> methanol> petroleum ether> acetone> chloroform> benzene. Mazumder *et al.*,⁴³ reported that methanol extract of leaves had high yield (11.47%) followed by petroleum ether (5.55%) and chloroform (3.06%). The yield was related to the weight of plant material used for extraction, geographic distribution and the environmental factors. Table (2) represents the yield of water-soluble extract and alcohol soluble extracts. High water-soluble extract yield of 14.15% was recorded while, the alcohol soluble extract yield was 12.50%. High water-soluble yield indicates that leaves are rich in carbohydrates, phenolic compounds and acids.

Water soluble and alcohol soluble yield of *A. adenophora* leaves reported by Negi *et al.*,⁴⁴ showed low percentage of yield in water soluble (13.71%) and alcohol soluble (11.84%) extracts. The inorganic composition and other substances (impurities) in a plant material can be determined with ash values. The total ash value was higher (10.31%) compared to acid insoluble (8.2%) and water soluble (4.1%) ash values (Table 3). The low acid insoluble ash value was because of siliceous matter in minute quantity. The ash value was consistent with the findings of Negi *et al.*,⁴⁴. The quantitative estimation of phytochemicals was evaluated to explore the plant's biological significance. *A. adenophora* leaves had a total phenol of 155.22 mg/g. The flavonoids were 134.85 mg/g, alkaloids and tannins content were 43.8 mg/g and 78.83 mg/g respectively (Table 4 and Fig. 2). The total phenol and flavonoid content of 30 mg/g and 510 mg/g respectively were reported in *A. adenophora*. Though the phenol content was low, the flavonoid content was high⁴⁵.

Table1: Percentage of yield obtained from different solvent extracts of *A. adenophora* leaves

S. No	Solvent	Colour of the yield	%of yield
1	Petroleumether	Darkgreen	2.495
2	Benzene	Black	1.215
3	Chloroform	Black	1.457
4	Acetone	Darkgreen	2.315
5	Methanol	Darkgreen	3.540
6	Aqueous	Darkgreen	4.105



Fig. 1. Fresh leaves of *Ageratina adenophora*

Table 2: Over all percentage of yield obtained from *A. adenophora* leaves

S. No	Extract	%ofyield
1	Watersolubleextract	14.15
2	Alcoholsolubleextract	12.50

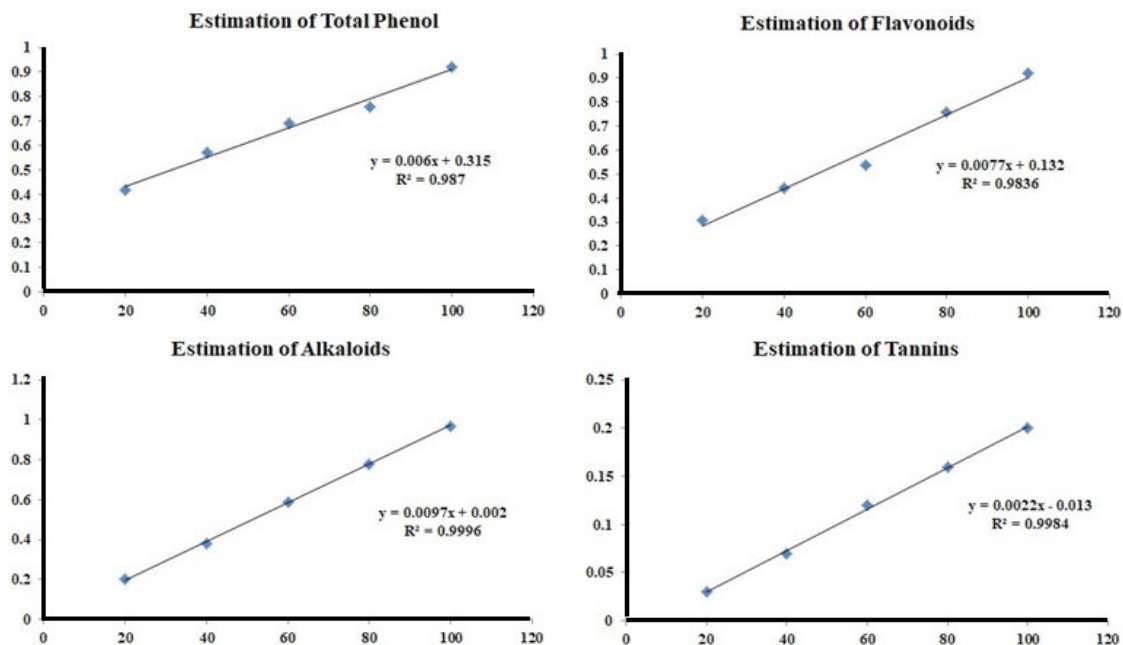


Fig. 2. Quantitative estimation of secondary metabolites in *A. adenophora* leaves

Table 3: Percentage of ash values obtained from *A. adenophora* leaves

S. No	Ash content	% of ash
1	Total ash	10.31
2	Acid insoluble ash	8.2
3	Water soluble ash	4.1
4	Loss due to drying	0.14

Table 4: Quantitative estimation of secondary metabolites in *A. adenophora* leaves

S. No	Phytochemicals	Quantity (mg/mL)	R ²
1	Phenol	155.22±2.66	0.9870
2	Flavonoids	134.85±0.36	0.9836
3	Alkaloids	43.88±0.73	0.9996
4	Tannins	78.83±1.25	0.9984

The qualitative estimation of aqueous leaf extract showed carbohydrates, glycosides,

flavonoids, steroids, tannins and terpenoids. Tannins are the only secondary metabolite that was not detected with petroleum ether. Similarly, glycosides, tannins and steroids were absent in chloroform extract. Alkaloids, steroids, tannins, and flavonoids were present in benzene extract. Glycosides, tannins, steroids and terpenoids were present in acetone extract. Methanol extract did not report for steroids (Table 5) Saponins in methanol extract⁴³ and tannins in petroleum ether⁴⁴ contradicted with this study. The phytochemicals aqueous extract as reported by Negi *et al.*,⁴⁴ was in concurrence with present finding. The difference in the phytochemicals as reported by previous investigation was because of geographical distribution of the plant and the phytochemicals indicates the pharmacological significance of the plant.

Table 5: Phytochemicals identified from *A. adenophora* leaves using different solvent extracts

Phytochemicals	Solvent extracts					
	Water	Pet. ether	Benzene	Chloroform	Acetone	Methanol
Carbohydrates	+	+	-	+	-	+
Proteins	+	+	-	+	-	+
Saponins	+	+	-	+	-	+
Alkaloids	+	+	+	+	-	+
Glycosides	+	+	-	-	+	+
Flavonoids	+	+	+	+	-	+
Terpenoids	+	+	-	+	+	+
Steroids	+	+	+	-	+	-

GC-MS analysis showed 20 different compounds (Fig. 3 and Table 6) with maximum proportion (76.17%) being occupied by four different compounds namely isolongifolene (22.73%), methyl ionone (22.26%), 2(3H) naphthalenone 4,4a,5,6,7,8-hexahydro-4a,7,7-trimethyl (R)- and D (-) quinic acid (14.20%). Isolongifolene is a polycyclic hydrocarbon and quinic acid is a polyphenol. 2(3H) naphthalenone 4,4a,5,6,7,8-hexahydro-4a,7,7-trimethyl (R)- and methyl ionone are sesquiterpenoids. Spathulenol, -bisabolol and 4,6,6, trimethyl-2-(3-methyl-buta-1,3-dienyl)-3-oxa-tricyclo [5.1.0.02,4] octane has been previous reported by Poudel *et al.*,¹⁸. Rangasamy and Namasivayam⁴⁶ reported that isolongifolene isolated from *Muuraya koenigii* exhibited effective *In vitro* antioxidant property via scavenging DPPH, ABTS, nitric oxide,

hydroxyl, and hydrogen peroxide super oxide free radicals. Quinic acid was reported for its antioxidant, hepatoprotective and anti-inflammatory activity⁴⁷. Bai *et al.*,⁴⁸ reported quinic acid with antibacterial activities. The chromatogram showed the major compounds were terpenoids and alcohol.

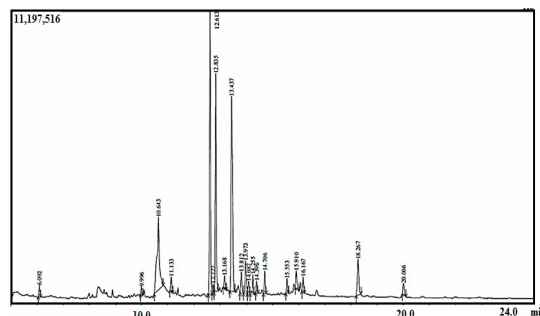


Fig. 3. GC-MS spectrum of *A. adenophora* leaves

Table 6: Identification of bioactive compounds of *A. adenophora* leaves

Peak	Area%	R. Time	M. formula	Compound name
1	0.69	6.092	C ₁₂ H ₂₀ O ₂	alpha-Fenchylacetate
2	0.608	9.996	C ₁₅ H ₂₄ O	(-)-Spathulenol
3	14.20	10.643	C ₇ H ₁₂ O ₆	D(-)-QuinicAcid
4	1.19	11.133	C ₁₅ H ₂₆ O	alpha-Bisabolol
5	22.26	12.612	C ₁₄ H ₂₂ O	Methylionone
6	0.73	12.727	C ₂₂ H ₃₄ O ₃	Kauran-18-al,17-(acetyloxy)-(4.beta.)-
7	16.98	12.835	C ₁₃ H ₂₀ O	2(3H)-Naphthalenone,4,4a,5,6,7,8-hexahydro-4a,7,7-trimethyl-,(R)-
8	1.00	13.168	C ₁₅ H ₂₄	Thujopsene
9	22.73	13.437	C ₁₅ H ₂₄	Isolongifolene
10	2.05	13.812	C ₁₅ H ₂₂ O	4,6,6-Trimethyl-2-(3-Methyl-Buta-1,3-Dienyl)-3-Oxa-Tricyclo[5.1.0.02,4]Octane
11	3.04	13.973	C ₁₆ H ₃₂ O ₂	Palmiticacid
12	1.21	14.087	C ₁₅ H ₂₄	Cedr-8-Ene
13	1.52	14.225	C ₁₆ H ₃₂ O ₂	Hexadecanoicacid
14	0.82	14.396	C ₁₅ H ₂₄	1-Methyl-4-Methylene-2-(2-Methyl-1-Propenyl)-1-Vinylcycloheptane
15	1.66	14.706	C ₁₅ H ₂₂ O ₂	6-(1-Hydroxymethyl-Vinyl)-4,8A-Dimethyl-3,5,6,7,8,8A-Hexahydro-1H-Naphthalen-2-One
16	1.10	15.553	C ₂₀ H ₄₀ O	Phytol
17	1.67	15.910	C ₁₈ H ₃₂ O ₂	Methylilinolenate
18	1.08	16.167	C ₂₀ H ₃₆ O ₂	Ethyllinoleate
19	4.24	18.267	C ₁₅ H ₂₄	Cycloheptan-4-Methylen-1-Methyl-2-(2-Methyl-1-Propen-1-yl)-1-Vinyl-(Humulen"-V
20	1.25	20.006	C ₂₀ H ₃₄ O	Thunbergol

Aa-AgNPs was synthesized with 5% plant extract (50 mL) and 3mM AgNO₃ (50 mL) was added with pH adjusted to 8.0. Plant extract when added to aqueous AgNO₃ altered the colour from light brown to greenish black (Fig. 4). It was then left in the dark for 18 h to ensure total nanoparticle saturation. FTIR identified the functional groupson Aa-AgNPs. The absorbance peaks at 3937.98 cm⁻¹, 3769.14 cm⁻¹, and 3206.64 cm⁻¹ in the FTIR spectrum (Fig. 5) showed OH-stretch of alcohol. CH-alkane and aldehyde were identified at 2974.18 cm⁻¹. The absorbance peak at 2901.59 cm⁻¹ corresponded to the C-H stretch of the methyl group. C=C aromatic alkanewas identified at 1581.21 cm⁻¹. Similarly,

the asymmetric stretch of a methyl molecule was identified at 1257.67 cm⁻¹. The peaks at 1030.42 cm⁻¹ and 799.57 cm⁻¹ respectively indicated the C-O stretch of alcohol and the C=C bend. Bromo and iodo compounds were represented with peaks at 687.94 cm⁻¹ and 571.67 cm⁻¹. Alcohol, aldehyde, alkane, methyl, aromatic amines, bromo and iodo chemicals contained in alkaloids, phenolic compounds, aminoacids, carbohydrates, and tannins were implicated in Aa-AgNPssynthesisand stability. The XRD investigation demonstrated the crystallinity of Aa-AgNPs (Fig. 6). The crystal plane displayed a significant peak at 2 angles of 38.23, 44.51, 64.41, and 77.51, which correspond to miller

indices of (111), (200), (220), and (311). (JCPDS No: 04-0783). These index planes confirmed the fabricated silver nanoparticles were crystalline and had a face-centred cubic structure. In addition, Scherrer equation analysis of peaks revealed an average particle size of 84nm. A diffraction spectrum for AgNPs mediated through *Andrographis paniculata* leaf extract was consistent with the XRD pattern found in this investigation.

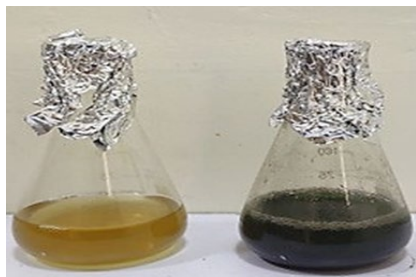


Fig. 4. *A.adenophora* leaf extract mediated silver nanoparticles(Aa-AgNPs) synthesis

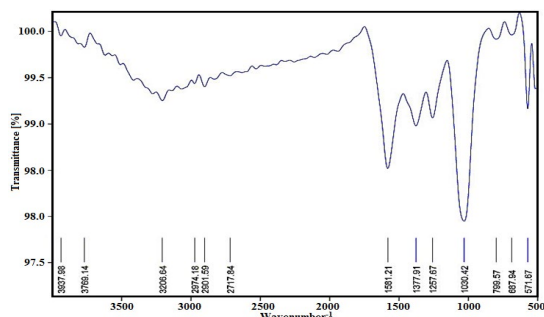


Fig. 5. FTIR absorbance spectrum of *A. adenophora* leaf extract mediated AgNPs

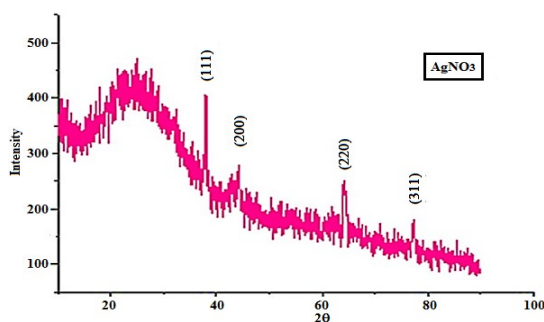


Fig. 6. XRD spectrum of Aa-AgNPs

SEM examinations validated the morphology, size, and form of the produced AgNPs. The SEM scans demonstrated that Aa-AgNPs were devoid of aggregation and were spherical (Fig. 7). AgNP synthesis from *Butea monosperma* flower extract showed spherical AgNPs²⁶. The antibacterial activity mediated by plant extracts depends on

phytochemicals⁴⁹. The antibacterial activity of Aa-AgNPs was due to surface coated phytochemicals and its synergistic activity with Aa-AgNPs. The phytochemical investigation of *A. adenophora* leaf extract showed alkaloids, saponins, steroids, tannins and terpenoids that are reported with antibacterial activities⁵⁰. The antibacterial activity of Aa-AgNPs (Fig. 8 and Table 7) was highly effective against gram negative bacterial species, exhibiting high bactericidal activity measuring 18mm against *K. pneumonia* and 16mm against *E. coli*. A decrease in the zone of inhibition against positive bacterial species, namely *S. pneumonia* and *S. aureus* measuring 15mm and 13mm was observed. The antibacterial potential of Aa-AgNPs may have been caused by the adhesion of AgNPs to the cell wall of bacteria, as suggested by the hypothesis. This binding of AgNPs to bacterial cell wall is caused by a difference in charge between the nanoparticles and the bacterial cell wall. Further, the high surface area of nanoparticles, size of the nanoparticles and the ability to generate free radicals might have influenced the effective antibacterial efficacy of nanoparticles^{51,52}. The binding of AgNPs to a bacterial cell wall stimulates conformational changes in membrane proteins, which results in an increase in membrane permeability. Subsequent AgNPs penetration causes cellular content leakage and cell death. The binding affinity of AgNPs towards sulphur and phosphorus in proteins and damaged DNA together contributes to its effective antibacterial potential. Das & Devkota⁵³, reported that aqueous extract of *A. adenophora* leaves showed bactericidal activity against *Klebsiella pneumonia*, *Enterococcus faecalis*, *Escherichia coli*, *Bacillus subtilis* and *Staphylococcus aureus*.



Fig. 7. SEM images of Aa-AgNPs

Table 7: Bactericidal activity of Aa-AgNPs

Bacterial species	Antibiotic Amoxicillin	Zone of Inhibition of AgNPs(mm)		
		40µg/mL	50µg/mL	60µg/mL
<i>Staphylococcus aureus</i>	9	12.5	12.5	13
<i>Streptococcus pneumoniae</i>	14	13.5	14	15
<i>Escherichia coli</i>	18.5	14	14.5	16
<i>Klebsiella pneumoniae</i>	15	15	16.5	18

Table 8: Antioxidant activity of Aa-AgNPs

Assay	Antioxidant activity of AgNPs				
	10 μ L	20 μ L	30 μ L	40 μ L	50 μ L
DPPH	35.00 \pm 2.55	43.89 \pm 3.12	59.25 \pm 3.29	66.31 \pm 2.07	76.90 \pm 2.59
ABTS	32.75 \pm 2.62	38.91 \pm 3.92	54.79 \pm 2.94	63.64 \pm 2.79	73.71 \pm 3.14
NO	38.72 \pm 1.02	43.84 \pm 1.88	64.41 \pm 3.71	73.45 \pm 3.50	83.21 \pm 2.53
Red.power	34.12 \pm 1.52	43.77 \pm 4.22	56.17 \pm 1.89	60.49 \pm 1.89	74.87 \pm 4.38

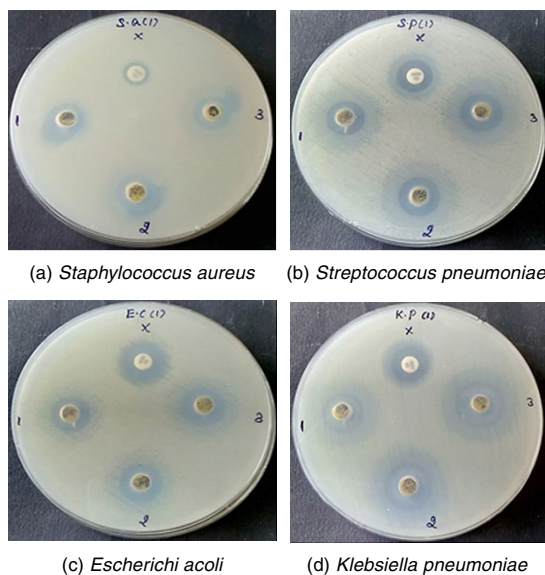
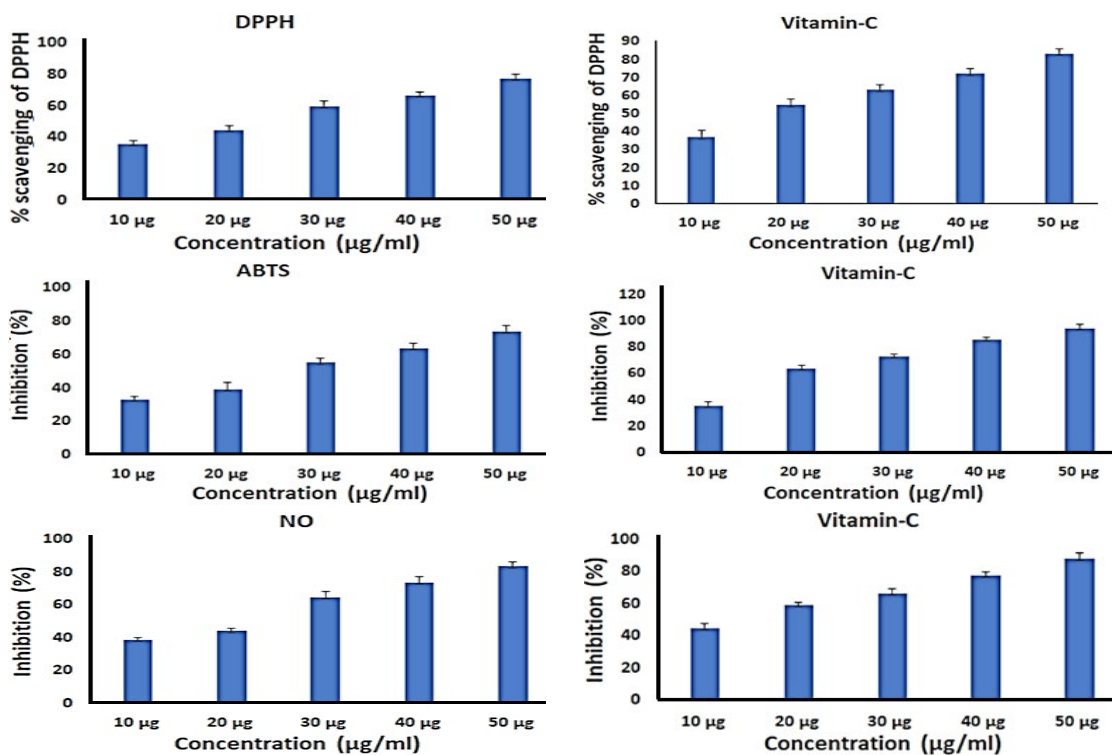


Fig. 8. Anti bacterial activity of Aa-AgNPs

Inhibition of DPPH activity of Aa-AgNPs was 76.90%, while ABTS radical inhibition was 73.71%. The Aa-AgNPs were effective in quenching nitric oxide (83.21%) and ferric reducing power (74.87%). The Aa-AgNPs were more effective in scavenging nitric oxide (Fig. 9 and Table 8). The methanol extract *A. adenophora* exhibited an IC₅₀ value of 92.791% against DPPH. Rajalakshmi *et al.*,⁵⁴ reported ethanol and aqueous extract of *A. adenophora* to be effective against nitric oxide. The results indicated that aqueous extract exhibited higher scavenging activity (55.16%) compared to ethanolic extract (40.48%). The reduction of ferric ions by aqueous extract was 339.97 mg/g and ethanolic extract was 326.48 mg/g⁵⁴. The effective scavenging of *A. adenophora* might be attributed to polyphenolic compounds and flavonoids. Also, the major compounds isolongifolone, quinic acid and methyl ionone, reported for their antioxidant activity might have played an effective role.



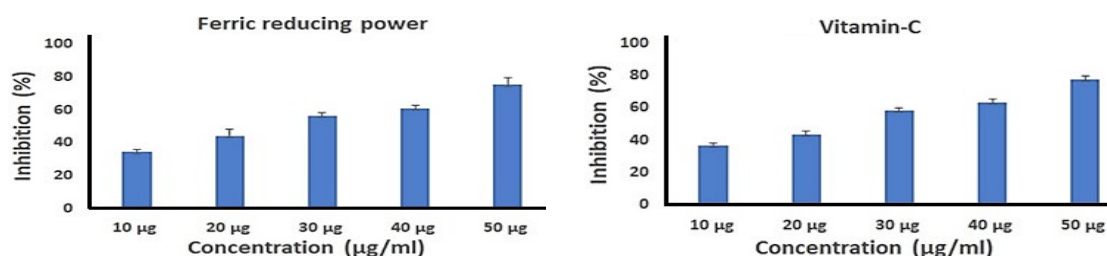


Fig. 9. Antioxidant activity of Aa-AgNPs

CONCLUSION

The present study revealed that the *A. Adenophora* leaves was rich in phytochemicals such as alkaloids, phenolic compounds, tannins, carbohydrates, proteins and aminoacids. The quantitative estimation of phytochemicals showed high amount of phenols, flavonoids, alkaloids and tannins in the leaves. Further, the phytochemicals are effective in the reducing metal ions in the synthesis of Aa-AgNPs and its stabilization. The

phytochemicals bound to surface of Aa-AgNPs might be attributed to their biological activity. Hence, *A. adenophora* might be a possible drug candidature and the study suggest further clinical trials before considering as a drug.

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