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Application Extended Vogel-Tammann-Fulcher Equation for soybean oil

IOANA STANCIU^{1*} and NOUREDDINE OUERFELLI²

 ¹ Department of Physical Chemistry, University of Bucharest, Faculty of Chemistry, 4-12 Elisabeta Blvd, 030018, Bucharest, Romania.
 ²Department Chemistry, College of Science, Imam Abdulrahman Bin Faisal University, P.O. Box 1982, Dammam 31441, Saudi Arabia.
 *Corresponding author E-mail: istanciu75@yahoo.com

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ABSTRACT

Mathematical models that describe the variation of soybean oil viscosity with temperature according to the recent WLF and VTF (or VFT) equations and traditionally by the Arrhenius equation. The Arrhenius equation shows that the viscosity of the oil is proportional to the absolute temperature and is determined by the activation energy parameter. In Arrhenius' equation the absolute temperature is replaced by T + b where both adjustable T and b are in °C. The mathematical models described by the equations WLF and VTF, are equal to each other. All three equations are in the same model when used for experimental data of temperature-viscosity dependence, they give exactly the same very high regression coefficient.

Keywords: Viscosity, Soybean oil, VTF Equation, Extend EVTF.

INTRODUCTION

The exact correlation of soybean oil viscosity data is of extreme practical importance when looking for optimal filler conditions for certain applications. Although the best correlations of experimental data for temperature dependence can be made by taking advantage of the VTF equation¹⁻³ and avoiding the use of temperature-dependent variables, correlations are still attempted by combining the use of both dependent variables, such as temperature as well as independent variables in many studies.

In previous research, the change in soybean oil viscosity at different temperatures was analyzed using absolute rate theory^{4,5}. Absolute rate theory, widely applicable to flow processes, describes the viscosity-temperature dependence in the form of Arrhenius^{5,9-6,7}. This theory has been used to determine changes in the viscoelastic properties of soybean oil at high temperatures^{6,8,9}. In this research, the most widely used empirical equation is the Williams-Landel-Ferry (WLF) equation¹⁰. Another equation widely used to model viscosity-temperature dependence is the Vogel-Tammann-Fulcher (VTF) equation¹⁻³. It was initially developed to analyze the

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viscosity-temperature relationship for overheated organic liquids and has recently been applied to polymers, protein solutions and foods¹¹⁻¹⁵.

In this paper, the viscosity data obtained at different temperatures were fitted and compared using both VTF and a proposed extended EVTF equation.

Viscosity temperature dependence Theoretical background

The fluid having a complex nature there is not yet a theory to describe it. There are some models in the literature such as the theory of the distribution function proposed by Kirkwood *et al.*,¹⁶, the molecular dynamics approach reported by Cumming and Evans¹⁷ and Eyring's reaction rate theory¹⁸⁻²⁰. Empirical and semi-empirical methods do not provide reasonable results, but they lack the general approach, especially in the vicinity of boiling temperature¹⁴.

Empirical equations

The dependence dynamic viscosity versus temperature absolute is described by several empirical equations.^{15,21-42}:

$$\ln \eta = A + \frac{B}{T+C} + a \log T + bT + cT^2 + \frac{D}{T^2} + \frac{E}{T^3} + \frac{F}{T^n}$$
(1)

Viscosity-temperature dependence for liquid systems that have a linear or nonlinear behavior, we represent the logarithm of the dynamic viscosity (lnη) in relation to the absolute temperature (1/T). Multi-constant equations (Eq.1) for many fluids that deviate strongly from Arrhenius behavior. There are several fluids such as melting salts, glasses and metals, ionic liquids, heavy and vegetable oils, fuels and biofuels, etc.¹⁶⁻¹⁸. For nonlinear behavior, it is found that the temperature dependence of the dynamic viscosity according to the Vogel-Fulcher-Tammann (VTF) equation¹⁻³ expressed as follows:

$$\ln \eta = \ln A_0 + \frac{A_1}{T - T_c}$$
(2)

Where A_0 and A_1 are optimal constants and Tc is the VTF temperature. It's also interesting to use the modified VTF equation which is expressed as follow:

$$\ln \eta = A_0 + \frac{E_1}{R(T - T_0)}$$
(3)

Where R-perfect gas constant, E_{τ} is the VTF activation energy, A_0 , T_0 , are identical to the A_0 and T_c parameters in Eq. (2), respectively.^{6,7-14-16}

Likewise, to describe the behavior of soybean oil at different temperatures we found in the literature the following relationship, named Willams-Landel-Ferry (WLF) equation and frequently used for honey⁴³⁻⁴⁵:

$$\ln(a_T) = \frac{-C_1(T - T_g)}{C_2 + (T - T_g)}$$
(4)

Material and methods

Non-additive soybean oil used in this study was produced in Romania. To determine the dynamic viscosity of soybean oil we used the Haake 550 viscometer with the VH1 viscose sensor at all shear rates. For temperature control we used a thermostatic bath at temperatures between 40 and 100°C.

RESULTS AND DISCUSSION

Table 1 shows the temperature range at which the determinations were made, the ln dynamic viscosity.

Temperature						ln(η/mPa.s)			
0C	К								
40	313.15	3.4484	3.2947	3.1655	3.0810	3.0573	3.0160	3.0146	3.0131
50	323.15	3.0015	2.7625	2.6575	2.5703	2.4732	2.4038	2.3712	2.3527
60	333.15	2.8693	2.6462	2.5848	2.5096	2.4292	2.3535	2.3243	2.2895
70	343.15	2.8003	2.5849	2.5112	2.4467	2.3749	2.3028	2.2670	2.2127
80	353.15	2.7389	2.5180	2.4354	2.3805	2.3253	2.2396	2.2148	2.1448
90	363.15	2.6899	2.4721	2.3749	2.3154	2.2690	2.1883	2.1389	2.0618
100	373.15	2.6484	2.4233	2.3504	2.2538	2.2083	2.1282	2.0894	2.0136

Table 1: Logarithm dynamic viscosity for soybean oil at different temperatures

Generally when the viscosity-temperature dependence deviates from the Arrhenius behavior^{21-23,25-30}, experimenters prefer to use the usual expression of VTF (Eq. 2) to minimize the discrepancy with experimental data.

If we extend the VTF model which has a linear dependence of the variables a second degree polynomial, we can write (Eq.5).

$$\ln \eta = \ln A_0 + \frac{A_1}{T - T_c} + \frac{A_2}{(T - T_c)^2} +$$
(5)

Where A_i are three new free adjustable parameters which can be determined with non-linear regression.



The Table 2 summarizes results of the

Fig. 1. Dependence of In dynamic viscosity versus absolute temperature at shear rate 3.3 s⁻¹ and comparison VTF model with EVTF proposed model



Fig. 2. Dependence of In dynamic viscosity versus absolute temperature at shear rate 6 s⁻¹ and comparison VTF model with EVTF proposed model

two different fits for VTF model (Eq. 2) and the present extended EVTF model (Eq. 5) related to the soybean oil system at seven different temperatures (293.15–353.15) K and for each fixed eight shear rates (3.3–120) s⁻¹. Overall, the *R*-square (R^2) and standard deviation (σ) are better for the proposed model (Eq. 5) when ($A_2 \neq 0$) than of the usual VTF model (Eq. 2) when ($A_2 = 0$) in Eq. 5.

Figures 1 to 8 show that the discrepancy between experimental values and the calculated ones by the EVTF proposed model (Eq. 5), in comparison with the simple VTF model (Eq. 2). In addition, the VTF model starts to deviate and diverge at high temperature.



Fig. 3. Dependence of In dynamic viscosity versus absolute temperature at shear rate 10.6 s⁻¹ and comparison VTF model with EVTF proposed model



Fig. 4. Dependence of In dynamic viscosity versus absolute temperature at shear rate 17.87 s⁻¹ and comparison VTF model with EVTF proposed model

3.2



Fig. 5. Dependence of Indynamic viscosity versus absolute temperature at shear rate 30 s⁻¹ and comparison VTF model with EVTF proposed model



Fig. 6. Dependence of In dynamic viscosity versus absolute temperature at shear rate 52.95 s⁻¹ and comparison VTF model with EVTF proposed model



Fig. 7. Dependence of In dynamic viscosity versus absolute temperature at shear rate 80 s⁻¹ and comparison VTF model with EVTF proposed model

Fig. 8. Dependence of In dynamic viscosity versus absolute temperature at shear rate 120 s⁻¹ and comparison VTF model with EVTF proposed model

Table 2: Optimal coefficients (A), VTF temperature (T_c), R-square (R^2) and standard deviation (σ), VTF
model (Eq. 3, $A_2 = 0$) and that for proposed EVTF model (Eq. 5, A2 $\neq 0$).

Model	Shear rate	Values of parameters				R-square	SD
	S ⁻¹	InA _o	A,/K	A_2/K^2	T_c/K	R^2	σ
VTF	3.3000	2.5220	11.074	0	301.4	0.99333	0.0225
EVTF		2.4501	15.338	-41.938	301.35	0.99832	0.0113
VTF	6.0000	2.3359	8.8138	0	303.1	0.99265	0.0255
EVTF		2.2649	12.522	-28.706	303.05	0.99818	0.0127
VTF	10.600	2.2570	8.8189	0	303.6	0.99062	0.0271
EVTF		2.1832	12.773	-32.265	303.56	0.99702	0.0153
VTF	17.870	2.1906	8.5447	0	304.7	0.98486	0.0339
EVTF		2.1189	12.354	-31.229	304.47	0.99124	0.0258
VTF	30.000	2.2001	4.9953	0	306.4	0.98327	0.0367
EVTF		2.1357	7.7854	-14.356	306.41	0.99060	0.0275
VTF	52.950	2.1082	5.5600	0	307.1	0.98591	0.0353
EVTF		2.0462	8.3008	-14.624	307.06	0.99180	0.0270
VTF	80.000	2.0735	5.4813	0	307.4	0.98369	0.0397
EVTF		2.0050	8.4498	-15.275	307.41	0.99056	0.0302
VTF	120.00	1.9611	7.8488	0	307.8	0.98595	0.0401
EVTF		1.8781	11.814	-25.307	307.47	0.99293	0.0283

Eqs. 2 and 5 can be re-expressed as follows:

$$ln\eta = lnA_0 + \frac{E_1}{R(T - T_c)}$$
(6)

for the VTF Model, and:

$$ln\eta = lnA_0 + \frac{E_1}{R(T - T_c)} - \frac{E_2^2}{R^2(T - T_c)^2} +$$
(7)

for the suggested extended EVTF model, where the parameters E_1 and E_2 are energies, *R* is the perfect gas constant, T_c and A_0 is a viscosity at infinite temperature.

The Table 3 summarizes values of the two new energies for the VTF model (Eq. 6) and the present extended EVTF model (Eq. 7) related to the soybean oil system at seven different temperatures (293.15–353.15) K and for each fixed eight shear rates (3.3–120) s⁻¹.

Table 3: Optimal Parameters E_1 and E_2 for at different shear rates, VTF model (Eq. 6, E2 = 0) and that for proposed EVTF model (Eq. 7, E2 \neq 0)

Model	Shear rate	Values of parameters					
		E,	E,	A_{o}			
	S ⁻¹	J.mol ⁻¹	J.mol ⁻¹	mPa.s			
VTF	3.30	92.074	0.0000	12.453			
EVTF		127.53	53.844	11.590			
VTF	6.00	73.282	0.0000	10.339			
EVTF		104.11	44.547	9.6302			
VTF	10.60	73.325	0.0000	9.5544			
EVTF		106.20	47.228	8.8747			
VTF	17.87	71.045	0.0000	8.9406			
EVTF		102.72	46.464	8.3220			
VTF	30.00	41.533	0.0000	9.0259			
EVTF		64.731	31.503	8.4630			
VTF	52.95	46.228	0.0000	8.2334			
EVTF		69.017	31.796	7.7384			
VTF	80.0	45.574	0.0000	7.9526			
EVTF		70.256	32.496	7.4261			
VTF	120.0	65.259	0.0000	7.1071			
EVTF		98.227	41.827	6.5411			

As a first step of attributing a significance meaning, (E_i) can be considered as the VTF activation energy, comparing with the mathematical Arrhenius-type equation. We note that the VTF temperature (T_c) is practically identical for the two models VTF and EVTF. We can justify this finding by the fact that the viscosity of soybean oil exponentially increases near a given temperature and physically diverges regardless of the used model such as VTF or EVTF. In addition, starting from that the viscosity value of solid state is almost infinite; we can conclude that the VTF temperature (T_c) is close and in relationship with the melting point (T_m). Comparing Eqs. 6 and 7, we can consider that (E_2) is a corrective term to the VTF activation energy (E_1) and must be highly correlated. By analogy with the Arrhenius-type equation, (A_o) is in causal correlation with the viscosity of the soybean oil at vapor state near the normal boiling temperature (T_b).^{15,46-49}

Effect of shear rate on the VTF-parameters.

Figure 9a shows that the pre-exponential factor (A_o) decreases exponentially at very low shear rate values, after that, it continues decreasing slowly. However, thinking about the physical significance of (A_o) which is equivalent to a viscosity and the eventual its obedience to the power law, we discover an optimal very interesting linearization revealed by the Fig. 9b, which permit us to suggest an interesting empirical model expressed as follows:

$$A_0 = \frac{a \cdot \gamma + b}{\gamma^{1.21}} \tag{8}$$

Where the 0.21-value can represent a certain rheological flow index and, (a) and (b) are adjustable parameters. Values of (a) and (b) are given into the (Figure 9b).



Fig. 9. Comparison of the logarithm of the pre-exponential factor In A_o (Table 2) calculated by Eq. (7), related to (·): VTF model (Eq. 3) and that (°): the proposed EVTF model (Eq. 5)

Figures 10 and 11 show that the optimal coefficients (A_p) decrease rapidly at low shear rate to reach a minimum and vary very slightly in the shear rate range between 30 and 80 s⁻¹. Considering that these parameters are in relationship with the VTF activation energy (E_p), we can conclude that there a certain stabilization in the³⁰⁻⁸⁰ shear rate range in (s⁻¹), where the molecules of soybean oil find facility to transit from one fluid layer to an adjacent one.



Fig. 10. Comparison of the optimal coefficients $(A_{,})$, (Table 2), related to (\cdot): VTF model (Eq. 3) and that ($^{\circ}$): the proposed EVTF model (Eq. 5)



Fig. 11. Variation of the optimal coefficients (A_{a}) , (Table 2), related to the proposed EVTF model (Eq. 5) as a function of the shear rate (Table 2)

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Figure 12 shows that the VTF temperature (T_c) increases with the shear rate and tends to a limiting value about 308 K at high values of shear rate which is probably in relationship with the melting point (T_m) .

CONCLUSION

The present paper proposes a new rheological model extending the usual Vogel-Tammann-Fulcher model (VTF) linear in $1/(T-T_c)$ to a non linear one (i.e. polynomial form in $1/(T-T_c)$) which can be called Extended Vogel-Tammann-Fulcher expression (EVTF) and depending on the discrepancy with experimental data, we can choose the optimal polynomial degree since two.

In addition, the present paper proposes a new EVTF model of dynamic temperature viscosity dependence for soybean oil. The present work comes into within the general framework of empirical and semi-empirical modeling by proposing original expressions in rheology or by extending certain existing models.^{42,50,51}

If we compare the activation energies of VTF (E_i), as well as the temperatures of VTF (T_c) to provide and classify the behavior of rheological fluids to provide some interesting theoretical interpretations and to contribute to the development of the theory.

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Conflict of Interest

No potential conflict of interest was reported by the co-authors.

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