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Growth and Characterization of Jet Nebulizer Spray Deposited n-type WO₃ thin films for Junction Diode Application

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ABSTRACT

The n-type tungsten oxide (WO $_3$) polycrystalline thin films have been prepared at an optimized parameters (0.20M, 5 ml and 500°C) using jet nebulizer spray pyrolysis (JNSP) technique. Such prepared WO $_3$ films were characterized by XRD, SEM, EDAX, UV-vis from I-V. The XRD pattern of the optimized WO $_3$ film reveals the monoclinic structure. The SEM and EDAX images shows that the surface morphological variations and elements present were confirmed. The optical properties were recorded by UV-vis spectrum and the maximum band gap value was observed as 3.86 eV for 500°C. The maximum conductivity of the prepared WO $_3$ was recorded as 1.201 x 10°8S/cm from I-V characterization for 500°C.Using J-V plot, the diode parameters of n-WO $_3$ /p-Si prepared at 500°C with 0.2 M and 5 ml were measured under dark and illumination. The ideality factor (n) and barrier height (Φ_b) values of n-WO $_3$ /p-Si diode are obtained as 5.8 and 0.80 eV in dark and 3.9 and 0.81 eV under illumination.

Keywords: Tungsten trioxidethin films, JNSP technique, P-N junction diode, Ideality factor, Barrier height.

INTRODUCTION

Transition Metal Oxides (TMOs) have attracted substantial interest owing to their potential applications such as sensors, solar cells, photocatalysis and electrochromic devices¹⁻⁵. ZnO,TiO₂, SnO₂, MoO₃ and WO₃ are the most standard semiconductor metal oxides used in the optoelectronic devices. Among them, tungsten

trioxide (WO₃) is used in many electronic devices because of its tunable properties of chemical composition, surface morphological, optical absorption in visible range and high the rmalstability^{6, 7}. Several preparation techniques have been used to deposit tungsten oxide thin films, including sol-gel, spray pyrolysis, chemical vapour deposition, RF sputtering, plasma evaporation, electrodeposition, etc⁸⁻¹⁸. In fact, spray pyrolysis is one of the most versatile techniques

to deposit oxide films, allowing the control of many properties of the films by changing the preparation parameters of the technique. It is very simple, economical and a suitable method for large area coating for many semiconducting thin films.

The aim of this work isto prepare WO_3 thin films using modified spray technique of jet nebulizer spray pyrolysis (JNSP) technique $^{19-21}$. The prepared WO_3 films were characterized by structural, optical and electrical properties. Finally, $\mathrm{n-WO}_3/\mathrm{p-Si}$ junction diode parameters were observed using the current-voltage (I-V) measurements.

EXPERIMENTAL

Preparation of WO, thin films

The precursor solution was prepared with different mole concentrations of 0.10, 0.15, 0.20 and 0.25 M of tungsten (VI) chloride(WCI_s) and de-ionized water. WO₃ thin films were coated on well cleaned glass substrates (2 x 2 cm) under three variable conditions to optimize deposition parameters as follows (i) Initially, the WO₃ thin films were prepared by varying the volume of solution (3, 5, 7 and 9 ml) with constantsubstrate temperature of 500°C and 0.10 mole concentration for optimizing the volume level of the solution. (ii) Then by varying the mole concentration (0.05, 0.10, 0.15 and 0.20 M) with fixed 500°C substrate temperature and volume of solution as 5 ml, for optimizing the mole concentration, (iii) Finally, WO. thin films were preparedby varying the substrate temperatures(400, 450,500 and 550°C) with optimized molarity as 0.2 M and volume level as 5 ml and the temperature is optimized.

Fabrication of n-WO₃/p-Si junction diode

Prior to the fabrication of junction diode, silicon wafer is cleaned well by Piranha solution (3:1 of $\rm H_2SO_4$: $\rm H_2O_2$) to remove impurities like organic residues, dust and grease. Next to remove the native oxides from the Si wafer by $\rm H_2O$: HF (10:1) solution. The precursor solution of 20 ml with 0.20 M was coated on p-type Si wafer(1 x 1 cm) at 500°C substrate temperature to form the n-WO₃/ p-Si junction diode. After the formation of n-WO₃/ p-Si diode, the silver paste (ELTECK Corporation) was applied on the both surfaces of

the junction for better electrical contacts. The Ag applied sample is dried at ambient condition for about 6 hours. The current-voltage (I-V) measurement under dark and light illumination (650 W halogen lamps) was carried out using Keithley Electrometer (6517B).

Characterization Technique

The X-ray diffractometer (XRD)(XPERT-PRO) with CuKα 1 radiation of wavelength 1.5406 Å at a generator setting of 30 m A and 40 kV in the 2θ range from 20° to 70° was used to obtain the structure of the WO₃ thin films. The surface morphology of the films were analyzed by the scanning electron microscope (SEM) (JEOL/ EOJSM-6390). The presence of elements (W and O) was confirmed by the energy dispersive analysis from X-ray spectroscopy (EDAX) (BRUKER). The UV-visible spectrophotometer (Perkin Elmer Lambda 35) showed the optical properties in the wavelength range from 300 to 900 nm. DC electrical properties and n-WO₃/p-Si junction diode characterizations was measured using Keithley Electrometer (6517-B).

RESULTS AND DISCUSSION

XRD Analysis

The deposited WO₃ thin films have been characterized by XRD analysis. Fig. 1a shows the XRD patterns of film deposited at different volume of solution from 3 to 9 ml with constant substrate temperature and mole concentration as 500°C and 0.10 M respectively. It shows that the films are polycrystalline. It also found that the deposited films at 3, 5, 7 and 9 ml are well crystallized and having monoclinic structure with major orientation along (022) plane. The preferred orientation remains along (022) plane for all films. The diffraction an angle (20 in degree) with corresponding hkl planes coincides with JCPDS values (Card no. 83-0950) and indicates monoclinic structure. Fig. 1b shows the XRD pattern of WO₃ for different mole concentrations 0.05, 0.10, 0.15 and 0.20 M with constant volume 5 ml and substrate temperature 500°C. The intensity of the diffraction peaks increases with increasing mole concentration from 0.05 to 0.20 M. The diffraction peaks confirm the monoclinic structure. Fig. 1c elucidates the XRD pattern of WO₃ for different substrate temperature from 400°C to 550°C for fixed 0.20 M and 5 ml. Fig. 1c reveals that the crystallinity increases as increasing the substrate temperature till 500°C and then decreases. It also confirms the monoclinic structure with preferred orientations of (022) planes.

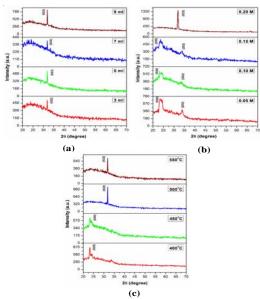


Fig. 1. XRD patterns for different (a) volume concentrations, (b) mole concentrations and (c) substrate temperatures.

The microstructural properties of crystallite size (D) and micro strain (ϵ) of WO $_3$ thin films for the preferred orientation of (0 2 2) were calculated by the following relations (1, 2) $^{20, 21}$.

$$\mathbf{D} = \frac{\mathbf{k}\lambda}{\mathbf{\beta}\mathbf{cos}\mathbf{\theta}} \tag{1}$$

$$\mathbf{\varepsilon} = \frac{\mathbf{\beta}\mathbf{cos}\mathbf{\theta}}{\mathbf{\theta}} \tag{2}$$

where k is the shape factor (k=0.94), λ is the wavelength of the X-ray radiation, θ is diffraction angle and β is the full width at half maximum.

Table 1illustrates the maximum crystallite size and minimum microstrain values for preferred orientation of (02 2) plane as 46.73 nm and 0.7887 x 10⁻³ lines⁻² m⁻⁴ (5 ml of volume). By varying mole concentration, the preferred orientation plane (2 0 2) changed to (0 2 2) for (Table 2) and maximum crystallite size and minimum microstrain values of 52.22 nm and 0.7086 x 10⁻³ lines⁻² m⁻⁴ for 0.20 M concentration were obtained. From varying substrate temperature result, the preferred orientation of (0 2 2) plane was not obtained for 400 and 450°C

owing to poor crystallinity in low temperatures. The substrate temperature of 500 and 550°C results the good crystalline nature and reveals the maximum crystallite size and minimum microstrain of 53.45 nm and 0.7878 x $10^{\text{-}3}$ lines $^{\text{-}2}$ m $^{\text{-}4}$ for the substrate temperature of 500°C (Table 3). The tables imply the variation of D and ϵ of WO $_{\!_3}$ films for various volumes of solution, mole concentrations and substrate temperatures. The results revealed that the 5 ml of volume, 0.20 M of concentration and 500°C of substrate temperature provide good crystalline nature of deposited films in the present study using JNSP technique.

Table. 1: Microstructural properties for different volume concentrations.

Volume variation (ml)	FWHM (Radians)	Crystallite size (D)nm	Microstrain (ε)(x 10 ⁻³ lines ⁻² m ⁻⁴)
3	0.0026	45.73	0.7887
5	0.0023	46.45	0.7567
7	0.0024	45.38	0.7672
9	0.0027	44.40	0.7856

Table. 2: Microstructural properties for different mole concentrations.

Molarity Variation (M)	FWHM (Radians)	Crystallite size (D)nm	Microstrain (ε)(x 10 ⁻³ lines ⁻² m ⁻⁴)
0.05 (2 0 2)	0.0128	43.75	0.8056
0.01 (2 0 2)	0.0115	44.70	0.7886
0.15 (2 0 2)	0.0108	49.40	0.7556
0.20 (0 2 2)	0.0085	52.22	0.7086

Table. 3: Microstructural properties for different substrate temperature.

	-	Microstrain (ε)(x 10 ⁻³ lines ⁻² m ⁻⁴)
0.0037 0.0033	53.45 44.03	0.6980 0.7878
	(Radians)	(Radians) size (D)nm 0.0037 53.45

SEM and EDAX Analysis

The surface morphological changes of ${\rm WO_3}$ films for various volume of solution, mole concentrations and substrate temperatures are shown in Fig. 2, 3 and 4. In Fig. 2 a & b, the closely packed sub-microsized and irregular shaped grains were observed for 3 and 5 ml of volume. When the volume of the solution is increased, the large variations in surface with randomly oriented sub-microsized rectangular shaped grains were observed (Fig. 2c & d). Figure.

3 a-d shows the surface morphology of different mole concentrations of WO_3 films. From the images its shows the randomly oriented submicrosized morphological results, which are similar to Fig. 2. The SEM images of WO_3 thin films of different substrate temperatures 400 to 550°C are shown in Fig. 4 a, d. It observes thatthe increase in substrate temperature grains are arranged sub-microsized

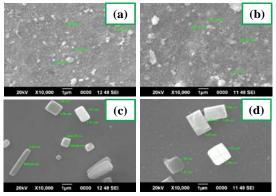


Fig. 2 SEM images for different volume concentrations of (a) 3, (b) 5, (c) 7 and (d) 9 ml.

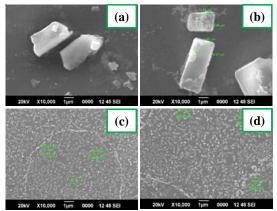


Fig. 3. SEM images for different mole concentrations of (a) 0.05, (b) 0.10, (c) 0.15and (d) 0.20 M.

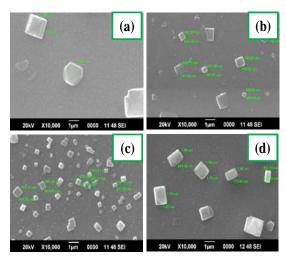


Fig. 4. SEM images for different substrate temperatures of (a) 400, (b) 450, (c) 500 and (d) 550°C.

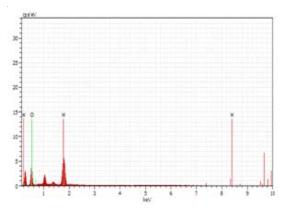


Fig. 5. EDAX spectrum of WO₃ thin film for the substrate temperature of 500 °C.

rectangular and square shaped structures of WO_3 films. Fig.5 depicts the EDAX spectrum of WO_3 thin filmsdepositedat a substrate temperatureof 500°C. It showed that the presence of elements tungsten and oxygen is confirmed.

Figure. 5. Depicts the EDAX spectrum of WO_3 thin film sdeposited at a substrate temperature of 500°C. It showed that the presence of elements tungsten and oxygen is confirmed.

UV-vis Spectra

Figure 6 a illuminates the variation of transmittance with wave length for the WO_3 thin films deposited at different substrate temperatures 400, 450, 500 and 550°C. The average transmission in the visible region has been found to vary from 70 to 85% depending upon the substrate temperature.

As increase in transmission is observed with increase in deposition temperature. At relatively lower temperatures less than 500°C, lower transmission is observed due to which is the formation of uneven films and incomplete decomposition of the sprayed droplets. The variation of transmittance owes to increase in packing density and decrease in thickness of the film which is due to shrinking of spray droplets²².

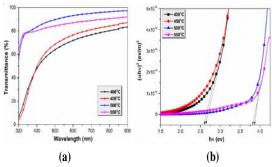


Fig. 6. (a) Transmittance spectra and (b) Band gap energy for different substrate temperatures.

From the UV-vis spectra, the band gap energy (E_g) for WO₃ films is calculated using equation (3)²³,

$$\alpha h v^2 = B(h v - E_g) \tag{3}$$

where α is the absorption co-efficient, ho is the photon energy, B is the constant and E is the band gap energy. Figure 6b displays the plot of (ahv)2 vs. (hv) for the WO3 thin films deposited at different substrate temperatures. It can be seen that the plots are linear in the region of strong absorption i.e. near fundamental absorption edge. Extrapolation of the linear portion to (ahv)2 is equal to zero gives the direct band gap energy of the films are found to vary as 2.63, 2.58, 3.86 and 3.80 eV for different substrate temperatures 400, 450, 500 and 550°C respectively. The maximum band gap energy 3.86 eV²⁴ is obtained for the substrate temperature 500 °C. This high value of band gap confirms the surface smoothness and uniformity of deposited thin films. Also a shift in band gap to higher energies is attributedto increased carrier density and to the change in crystallinity of the films with temperature²⁵.

DC Electrical Properties

The current and voltage (I-V) measurements of WO_3 thin films prepared for various volume of solution, mole concentrations and

substrate temperatures were taken at room temperature using Keithley Electrometer through two probe. The current values were measured for different applied voltages from 10 to 100 V. The DC conductivity (σ) for WO₃ films was calculated using the given equation(4)^{20,21},

$$\sigma = \left(\frac{1}{V}\right) \times \left(\frac{d}{A}\right) \tag{4}$$

Where I is current, V is applied potential, d is inter-probe distance and A is cross-sectional area of the film.

Figure.7a shows the variation of conductivity with different volume of solution of deposited WO3 thin films. It shows that the average conductivity values are 3.443 x 10⁻⁸, 5.133 x 10⁻⁸, 4.395 x 10-9 and 3.699 x 10-9S/cm for volume of solution from 3 to 9 ml respectively. Also shows that the conductivity value increased up to 5 ml and then decreases. Fig. 7b shows the average conductivity of deposited WO3 thin films at different mole concentrations of 0.05, 0.10, 0.15 and 0.20 M. The graph shows that the maximum conductivity 1.631 x 10⁻⁸ S/cm for the film deposited at 0.20 M. The variation of average conductivity with different substrate temperature of deposited WO3 thin films as shown in Fig. 7c. It observed that the average conductivity values of WO₃ films are 4.349 x 10⁻⁹, 5.726 x 10⁻⁹, 1.201 x 10⁻⁸ and 1.130 x 10⁻⁸S/cm. Also shows that the film deposited at substrate temperature of 500 °C have the maximum value of conductivity. The variations of values in conductivity may be due to the oxygen vacancies in the prepared

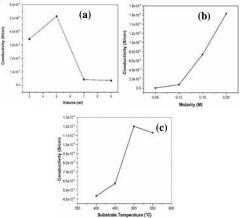


Fig. 7. Average conductivity values for different (a) voume concentrations, (b)mole concentrations and (c) substrate temperatures.

 ${
m WO}_3$ thin films^{21, 26-28}. The increasing of conductivity with the substrate temperature owing to the variation in morphology and the increasing of grain size, which reduced the lattice dislocations and imperfections of the W-O matrix. This phenomenon decreases the grain boundary volume associated with flow of charge carriers²⁹.

I-V characterization of n-WO₃/p-Si junction diode

The junction formation of the P-N diode is carried out by the n-type WO3 and p-type Si substrate. The P-N junction diode of n-WO_/p-Si was prepared by the optimized conditions of volume of solution (5 ml), mole concentration (0.20 M) and substrate temperature (500°C) by JNSP technique. The prepared n-WO₂/p-Si junction diode parameters of ideality factor (n) and barrier height $(\Phi_{\mathbf{k}})$ were measured under darkness and illumination of light (halogen+metal halide-100 mW/ cm²)²⁰. The reverse to forward bias current measurement is taken for the applied voltage of -4 to +4 V. Fig. 8 shows the I-V characteristics of n-WO₂/p-Si junction diode under dark and illumination. The n-WO₂/p-Si junction diode exhibits a good rectifying nature in dark and under illumination.

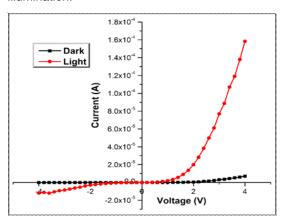


Fig. 8. I-V characterization for n-WO₃/p-Si junction diode in darkness and under illumination

Figure. 9 illustrates the semi-log plot of the current density (ln J) versus applied voltage (V). The diode parameters of ideality factor (n) and barrier height ($\Phi_{\rm b}$) are calculated from the forward bias J-V plot. These parameters provide an information about the charge transport mechanism of the p-n junction interface and the electrode contact with the tungsten oxide layer. Using the thermionic emission (TE) equations, the current

density of the n-WO $_3$ /p-Si junction diode was calculated as follows (5) 21,30 ,\

Where J_{\circ} is reverse saturation current density, q is charge of electron, V is applied potential, n is ideality factor, K is Boltzmann constant and T is absolute temperature.

Using equation (5) for V \geq 3kT/q, n and J $_{\circ}$ values were obtained from the slope and interception of semi-log forward bias J-V plot. Thus, n and $\Phi_{\rm b}$ were calculated by the following equations (6,7) 21,30 ,

$$J = J_0 \exp\left(\frac{qV}{nKT} - 1\right) \tag{5}$$

$$n = \frac{q}{KT} \frac{dV}{d(\ln J)} \tag{6}$$

$$\Phi_{b} = \frac{KT}{q} \ln \left(\frac{AA^*T^2}{J_0} \right) \tag{7}$$

Where A is the active area of prepared diode and A' is the Richardson constant.

The n and $\Phi_{\rm b}$ values under darkness is obtained as 5.8 and 0.80 eV and under light illumination as 3.9 and 0.81 eV. The n value is unity (i.e., n = 1) for an ideal P-N diode but in the present work, the n values are obtained as more than unity. The obtained result of n value is comparable to the reported value of 3.5, for a reactive magnetron sputtering deposited Ag/n-WO₃/p-Si Schottky diode³¹. The results suggest that the non-ideal behavior

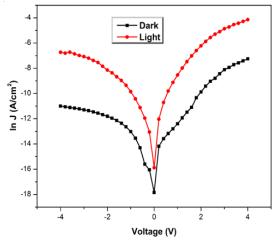


Fig. 9. Plots In J vs. V for n-WO₃/p-Si junction diode in darkness and under illumination.

of prepared n-WO₃/p-Si junction diode which may be owing to the presence of an interfacial thin native oxide layer (SiO₂) and barrier in-homogeneities^{20,21,30}. It may also be owing to series resistance and nonlinear metal-semiconductor contact^{20,32}. Another reasons may be the abnormalities of the inorganic film thickness and non-uniformity of the interfacial charges^{20,33}.

CONCLUSION

The $\mathrm{WO_3}$ thin films have been prepared using JNSP technique and the conditions of volume of solution, mole concentrations and substrate temperatures were optimized. From the XRD results, it was observed that the prepared $\mathrm{WO_3}$ film has monoclinic structure and also has the maximum crystallite size as 46.45, 52.22 and 53.45 nm for 5 ml of volume, 0.20 M of molarity and 500°C of substrate temperature respectively. The surface morphological variations of sub-microsized grains of $\mathrm{WO_3}$ were displayed by SEM analysis for different volume, molarity and substrate temperatures and the presence of elements such as tungsten and

oxgen were confirmed by EDAX results. UV-vis for various substrate temperatures was observed and it showed the maximum transmissionvalue and maximum band gap energy of 3.86 eV for the substrate temperature of 500°C. The DC electrical conductivity reveals that the 5 ml of volume, 0.20 M of molarity and 500°C of substrate temperature has maximum conductivity values of 5.133x10⁻⁸, 1.631x10⁻⁸ and 1.201x10⁻⁸ S/cm. Based on the optimized volume (5ml), molarity (0.20 M) and substrate temperature (500°C), the n-WO₃/p-Si junction diode was prepared using JNSP technique. The diode parameters of ideality factor (n) values were measured as 5.8 in darkness and under illumination as 3.9. The photoconduction nature of the prepared n-WO₃/p-Si junction diode will be endorsed for photo-detector application.

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