



Determination of Chlorpyrifos in Fruits and Vegetables by using Analytical Techniques

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ABSTRACT

Highly sensitive spectrophotometric and gas chromatographic methods were developed for the determination of chlorpyrifos. Alkaline hydrolysis of chlorpyrifos to 1,2,4-trichloropyridine was the basis followed by coupling with Congo red in presence of nitric acid. The absorption maxima of blue color formed at 80^o-90^oC was formed at 605nm. Beer's law was obeyed in the range of 0.5-5.7ppm. The standard deviation was found to be ± 0.005 . The method was free from other pesticide interferences. The method was applied to various fruits and vegetables procured at various agricultural fields at Sabbavaram area, Visakhapatnam, India and found satisfactory. Keeping in view of extensive use of pesticides on fruits and vegetables damaging the ecological balance. Determination of pesticide residue in the various vegetables collected are cauliflower, potato, spinach, and fruits such as pink apple, grapes etc were analysed by using Gas chromatography with electron capture detector (GC-ECD). Residues were extracted with hexane, dichloromethane and ethyl acetate (1:1% V/V). Clean up procedure was carried by applying solid phase extraction column. The recoveries of chlorpyrifos pesticide in fruits and vegetables selected were in the range of 80 to 98% fortified at 0.5mg/Kg. Some grape samples showed above the limit of quantification.

Keywords: Chlorpyrifos, Congo red, Spectrophotometer, GC-ECD, Fruits, Vegetables.

INTRODUCTION

Chlorpyrifos is a crystalline organophosphate insecticide. The IUPAC name of chlorpyrifos is *O*, *O*-diethyl *O*-3,5,6-trichloro-2-pyridyl phosphorothioate and with molecular formula C₉H₁₁Cl₃NO₃PS. Chlorpyrifos is moderately toxic and chronic exposure has been linked to neurological effects, developmental disorders, and autoimmune disorders. Chlorpyrifos is manufactured by reacting 3,5,6-trichloro-2-pyridinol with diethylthiophosphoryl chloride¹. Chlorpyrifos

is registered only for agricultural use, where it is "one of the most widely used organophosphate insecticides", according to the United States Environmental Protection Agency (EPA). The crops with the most intense chlorpyrifos use are cotton, corn, almonds, and fruit trees including oranges and apples. It is produced via a multistep synthesis from 3-methylpyridine.

Chlorpyrifos is an organophosphate, with potential for both acute toxicity at larger amounts and neurological effects in fetuses and children

even at very small amounts. Recent research indicates that children exposed to chlorpyrifos while in the womb have an increased risk of delays in mental and motor development at age 3 and an increased occurrence of pervasive developmental disorders such as ADHD². An earlier study demonstrated a correlation between prenatal chlorpyrifos exposure and lower weight and smaller head circumference at birth³. A study of the effects of chlorpyrifos on humans exposed over time showed that people exposed to high levels have autoimmune antibodies that are common in people with autoimmune disorders. There is a strong correlation to chronic illness associated with autoimmune disorders after exposure to chlorpyrifos⁴.

The immediate health hazard from air born chlorpyrifos in the examined houses was negligible, but the findings suggest that it is necessary to monitor chemicals which may contaminate indoor air and to assess the risk of prolonged exposure to such chemicals. The measuring of urinary metabolite TCP of chlorpyrifos via biological monitoring would be useful, allowing comprehensive evaluation of the exposure to chlorpyrifos in indoor air⁵. Chlorpyrifos is used for termite control in construction, forestry and field crops.

Numerous instrumental methods have been described for the detection/determination of chlorpyrifos generally analysed by spectrophotometry⁶⁻⁷, thin layer chromatography (TLC)⁸, and GC-MS⁹, liquid chromatography, mass spectrometry¹⁰, and Gas chromatography¹¹. The purpose of this study was to develop a spectrophotometric method and an analysis scheme for determination of chlorpyrifos in cauliflower, potato, spinach, and fruits such as pink apple, grapes etc by GC-ECD.

In this paper the author extracted endosulfan pesticide in fruits and vegetables using hexane, hexane-diehtyl ether, hexane and dichloromethane as solvents, clean up on solid phase extraction (SPE) and determined on Gas Chromatography with electron capture detector (GC-ECD).

EXPERIMENTAL

Chemical and reagents

A JASCO (Model UVIDEC-610 UV-VIS Spectrophotometry with 1cm matched quartz cuvettes) was used for all absorbance measurements. Systonics P^Hmeter (model 331) is used. All chemicals and reagents used are AnalR grade. The organic solvents like dichloromethane, ethyl acetate and hexane used were HPLC grade and purchased from E Merck. Technical grade pesticide standards were used for standardisations. Anhydrous sodium sulphate (AR) from E Merck used for residue extraction.

Extraction and clean up

Each vegetable was chopped into small pieces. All vegetables and fruits were collected from agricultural fields near Sabbavaram area, Visakhapatnam District, India. A representative sample (50gm) was taken with 5-10gm anhydrous sodium sulphate in blending machine to make fine paste. The sample was extracted with 100 ml of ethyl acetate, hexane or dichloromethane on mechanical shaker for one hour, extract was filtered, concentrated up to 5 ml on rotary evaporator and finally injected into GC-ECD. The clean-up of chlorpyrifos was performed out by using column chromatography. Packed with Florisil. The extract was eluted with ethyl acetate, hexane or dichloromethane. Elute was concentrated to 5-10 ml on rotary evaporator.

Sampling

Several samples of Water, Fruits and Vegetables were collected from Agricultural fields in Sabbavaram, Visakhapatnam district. Samples of one kilogram Grapes, pine apple (fruits) and Carrot, Cabbage, (vegetables) were procured and kept in refrigerator till analysis.

Procedures

Spectrophotometric method

1mL of alcoholic potassium hydroxide was added to an aliquot of working standard of chlorpyrifos (0.5-6.0 µg/mL). 2.0mL of 0.1% Congo red and 2mL of 1:1 nitric acid are added to give pale blue color. The solution was kept aside for 5 min before taking absorbance and absorbance

was measured at 605 nm against reagent blank. The absorbance corresponding to the bleached color which in turn corresponds to the analyte chloropyrifos concentration was obtained by subtracting the absorbance of the blank solution from that of test solution.

Gas Chromatography Instrumentaion

The analytical method was standardized by processing spiked samples in triplicate. Control samples were processed along with spiked ones. The details are given in Table 1.

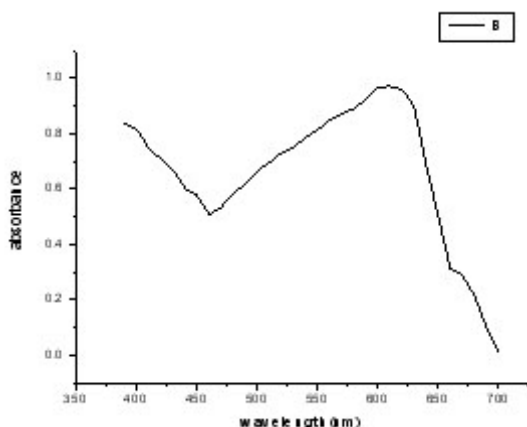


Fig 1: Absorption maxima of chloropyrifos

RESULTS AND DISCUSSION

The method is based on the alcoholic alkaline hydrolysis with potassium hydroxide. The reaction with potassium hydroxide in presence of an acid reagent forms an orange yellow color and measured at 605 nm. The absorption maximum is shown in Fig 1. The decrease tendency in absorbance is proportional to endosulfan. The Beer's law is obeyed in the range of 0.5-11 $\mu\text{g/mL}$. The consumption pattern of pesticides from agricultural source¹² is given in Fig 2.

Presence of pesticide residues in water, fruits and vegetables has become a global phenomenon. For the development of solid phase extraction conditions, a volatile solvent system must be used, as rapid evaporation of a large volume would be required in sample preparation without causing loss of volatile pesticides. The solvent system must be sufficiently polar to extract most polar pesticides. A flow rate of 0.5 mL/min was sufficient to recover all the pesticides except hexaconazole. It was noted that the solid phase extraction should not be left dry after conditioning. This could result in a significant loss of pesticides. It was observed that increasing the polarity of solvent gives lower recoveries. Twelve samples of Water, Fruits and Vegetables were collected from Agricultural fields in Sabbavaram, Visakhapatnam district for the determination of chloropyrifos and are tabulated in Table 2.



Fig. 2: Consumption pattern of pesticides Proposed scheme

Proposed Scheme

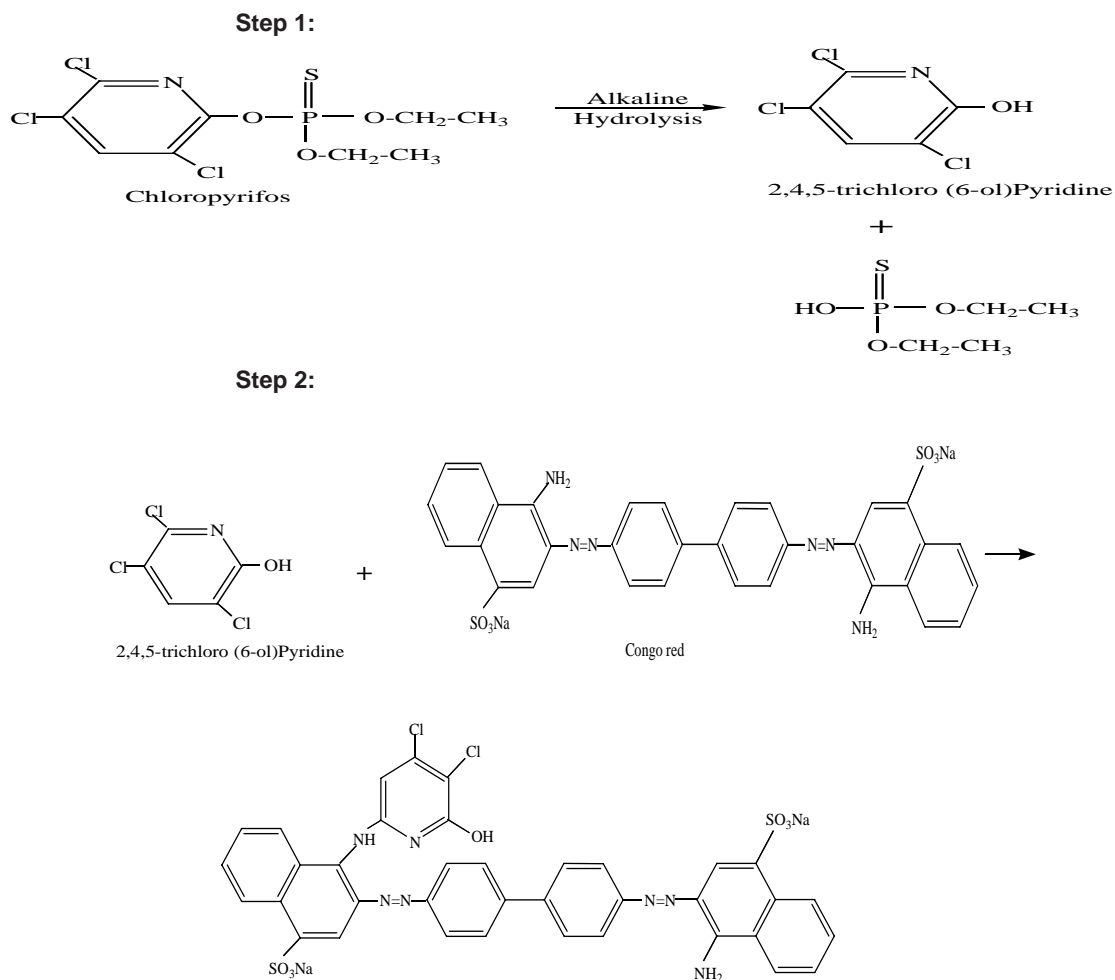


Table 1: Instrumental Parameters

S.No	Parameters	Make	Specifications
1.	Gas Chromatography	AGILENT-6820	
2.	Detector	ECD	
3.	Carrier gas	Nitrogen	99.99%
4.	Solvent	Hexane	HPLC grade
5.	Injection volume	Hamilton	0.5 mic.L
6.	column	Intercap	30M length, 0.25 mmID, 0.25 mic DF
7.	Oven temperature programming	Endosulfan	3 min 70.c, 2-25 min 108.c

Table 2 :Chloropyrifos in water, fruits and vegetables in sabbavaram, Visakhapatnam district

S.no	Name of the item	Pesticide residue, mg/Kg
1.	Water	0.30,0.29,0.27
	Banana field	2.12,2.06,2.08
2.	Fruits	Pine-apple 2.86,2.71,2.39
	Grapes	5.62,5.12,5.09
3.	Vegetables	Carrot 5.59,5.11,4.99
	Cabbage	5.60,5.76,5.14

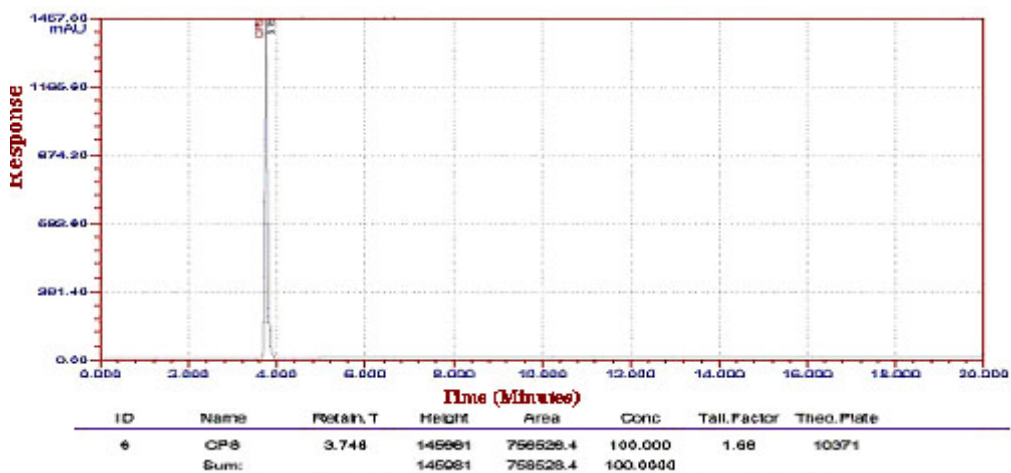


Fig. 3: GC Chromatogram of Cabbage extract after clean up

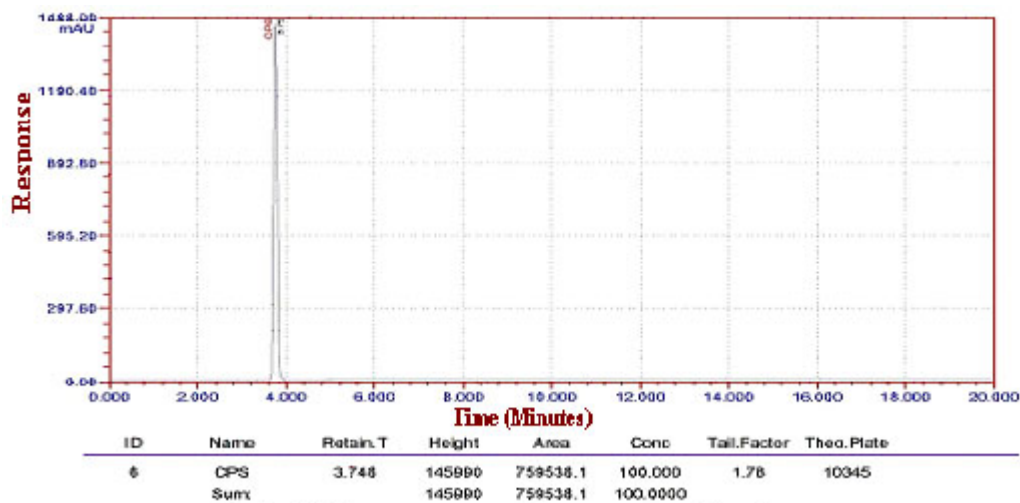


Fig. 4: GC Chromatogram of Carrot extract after clean up

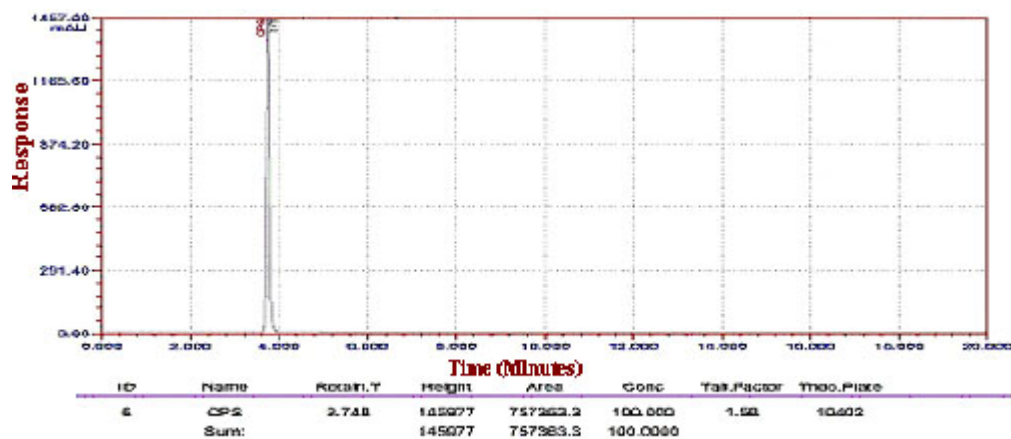


Fig. 5: GC Chromatogram of Pink Apple extract after clean up

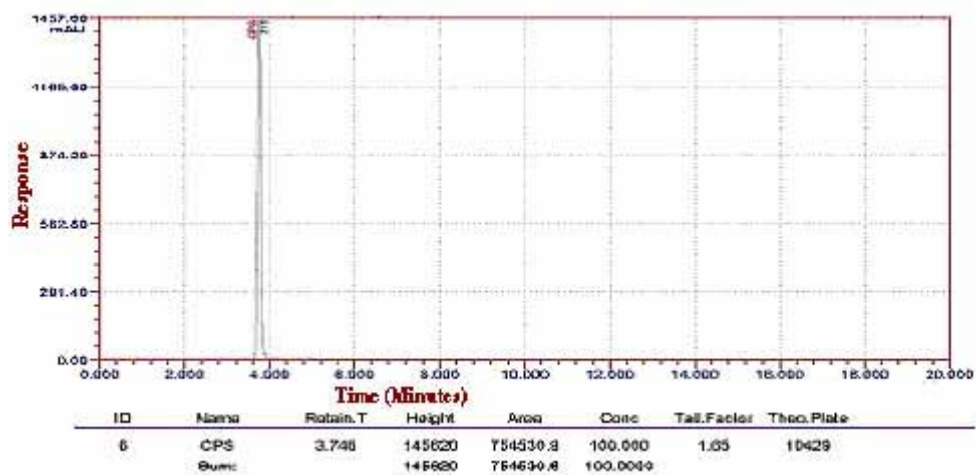


Fig. 6: GC Chromatogram of Grape extract after clean up

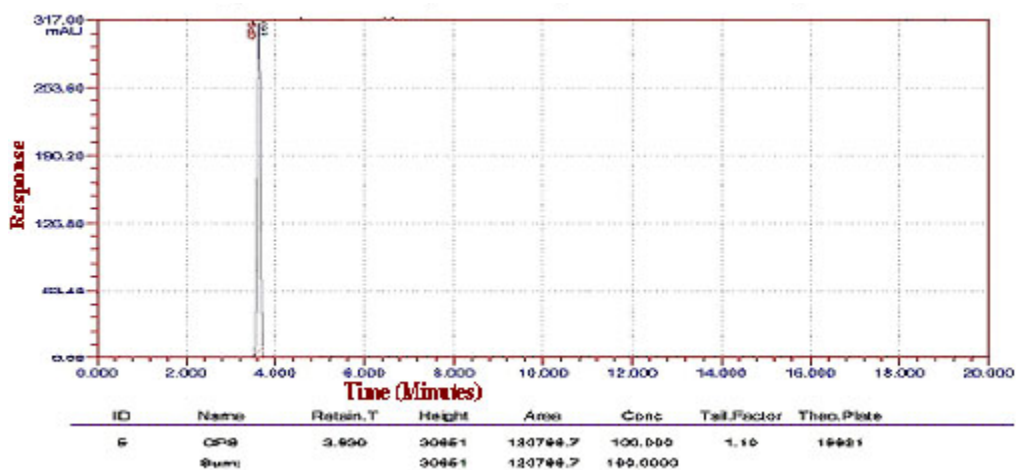


Fig. 7: GC Chromatogram of Chili field water extract after clean up

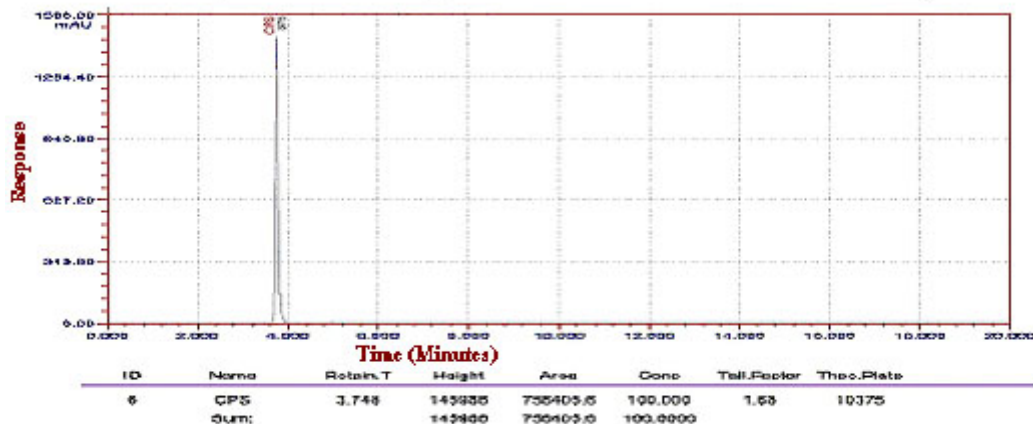


Fig. 8: GC Chromatogram of Banana field water extract after clean up

The present method was tailor-made in view of the previous information about the most prevalent pesticides in the area. From the results it can be seen that chlorpyrifos was detected in the range of 0.27-5.62 mg/ kg. The recoveries for chlorpyrifos in various samples ranged from 75-98% with coefficient of variance 1.3-7.6%. The various chromatograms for the determination of endosulfan were given in Fig 3-8.

CONCLUSION

GC-ECD has great potential for determining organochlorine pesticides in food extracts at low levels. In general, chlorpyrifos residues in food are well below the tolerance levels established for various food types by the FAO/WHO.

The results from this study showed that recoveries obtained were comparable to those obtained from the established silica gel cleanup method, However lower recoveries occasionally occurred for chlorpyrifos pesticide in water, fruits and vegetables using both cleanup methods. The benefits of the solid phase extraction method are less solvent consumption, no hazardous solvent is used, no cross-contamination and shorter analysis time.

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