



A Multistep Preparation of 3-Aryl-8-methoxythiazolo [3', 2' : 2, 3] [1, 2, 4] triazino [5, 6-b] indoles Under Microwave IR-radiations

RAVINDER SINGH

Department of Chemistry, Government College for Women, Sampla, India.

*Corresponding author E-mail: gahlawat.ravinder@gmail.com

<http://dx.doi.org/10.13005/ojc/300141>

(Received: October 16, 2013; Accepted: November 11, 2013)

ABSTRACT

The different 3-Aryl-8-methoxythiazolo [3', 2' : 2, 3] [1, 2, 4] triazino [5, 6-b] indoles having antihistaminic, antithyroid, antitubercular, antifungal & antibacterial activities are synthesized through a multistep preparation in high yield in shorter reaction time under microwave irradiations.

Key words: Indole, Microwave, Aryl, heterocyclic.

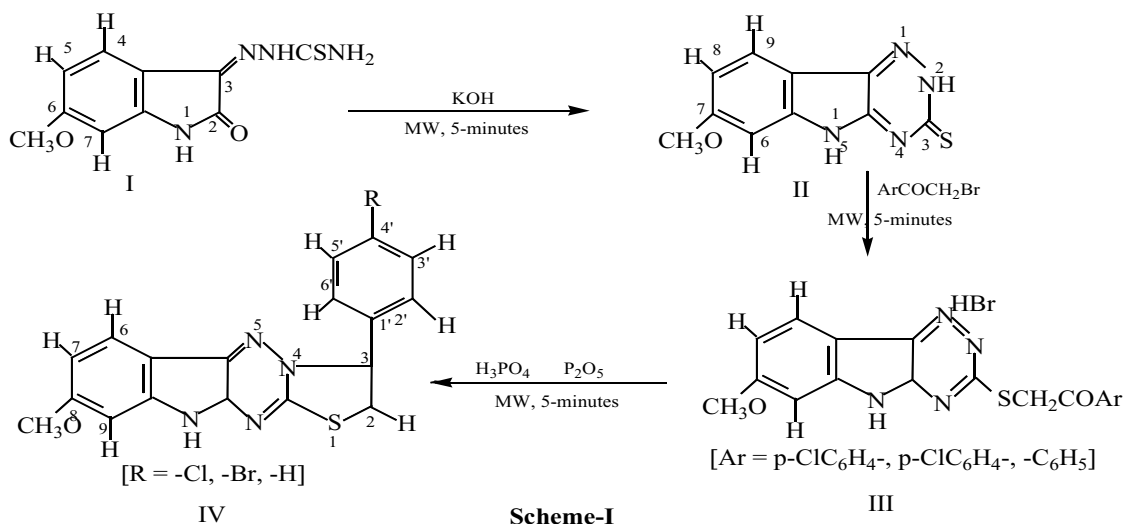
INTRODUCTION

Bridgehead nitrogen heterocycles containing thiazole and related heterocycles (thiadiazole and thiadiazine) exhibit antihistaminic, antithyroid, antitubercular, antifungal & antibacterial activities¹⁻³ and their synthetic importance has been greatly enhanced by the recent uses of their condensed bridgehead nitrogen heterocycles as anthelmintics, antidepressants, platelet aggregation inhibitors, antineoplastic, vulcanization accelerators and photographic sensitizers⁴⁻¹¹. The indoles are already been synthesized by different method But they requires longer reaction time and tedious workup¹²⁻²¹. Microwave assisted reactions are gaining much more importance in synthetic organic chemistry due to dramatic reduction in time from days to hours and hours to minutes or seconds²²⁻²⁴.

The present work reports the synthesis of 3-Aryl-8-methoxythiazolo[3', 2' : 2, 3][1, 2, 4] triazino[5, 6-b]indoles in a multi step preparation in high yield in shorter reaction time(Scheme 1).

Our work started by reacting 6-methoxyisatin with thiosemicarbazide in Anhyd. ethanol under microwave irradiation at 560W for 5-minutes to give 6-Methoxyisatin-3-thiosemicarbanzone (I). After separation, the 6-Methoxyisatin-3-thiosemicarbanzone(I) reacts with 5% KOH under microwave irradiation at 560W for 5-minutes to give 7-Methoxy-5H-2,3-dihydro [1, 2, 4] triazino [5, 6-b] indole-3-thione (II).

The compound 7-Methoxy-5H-2,3-dihydro [1, 2, 4] triazino [5, 6-b] indole-3-thione (II) further reacts with p-chlorophenacyl bromide under



microwave irradiation at 560W for 5-minutes to give 5H-3-(p-chlorophenacylthio-7-methoxy [1, 2, 4] triazino [5, 6-b] indole hydrobromide (IIIa; Ar = p-C₁C₆H₄). Similarly, 7-Methoxy-5H-2,3-dihydro [1, 2, 4] triazino [5, 6-b] indole-3-thione(II) was also irradiated with p-chlorophenacyl bromide, phenacyl bromide, under microwave irradiation at 560W for 5-minutes to give 5H-3-(p-bromophenacylthio-7-methoxy [1, 2, 4] triazino [5, 6-b] indole hydrobromide (IIIb, Ar = p-Br C₆H₄) and 5H-3-(phenacylthio-7-methoxy [1, 2, 4] triazino [5, 6-b] indole hydrobromide (IIIa; Ar = C₆H₅) respectively. The results are shown in Table-1.

We further explore our work by irradiated 5H-3-(p-chlorophenacylthio-7-methoxy [1, 2, 4] triazino [5, 6-b] indole hydrobromide (IIIa; Ar = p-C₁C₆H₄) in a mixture of H₃PO₄/P₂O₅ under microwave irradiation at 560W for 5-minutes to give 3-(p-chlorophenyl-8-methoxythiazolo[3', 2': 2, 3][1, 2, 4] triazino[5, 6-b]indole (IVa, R = Cl). Similarly, 5H-3-(p-bromophenacylthio-7-methoxy [1, 2, 4] triazino [5, 6-b] indole hydrobromide (IIIb; Ar = p-BrC₆H₄) and 5H-3-(phenacylthio-7-methoxy [1, 2, 4] triazino [5, 6-b] indole hydrobromide (IIIa; Ar = C₆H₅) were also irradiated in a mixture of H₃PO₄/P₂O₅ under microwave irradiation at 560W for 5-minutes to give 3-(p-bromophenyl-8-methoxythiazolo[3', 2': 2, 3][1, 2, 4] triazino[5, 6-b]indole (IVa, R = Br) and 3-(phenyl-8-methoxythiazolo[3', 2': 2, 3][1, 2, 4] triazino[5, 6-b] indole (IVa, R = H) respectively. The results are shown in Table-2.

EXPERIMENTAL

All the melting points reported are uncorrected. Infrared spectra (ν_{\max} in cm⁻¹) were recorded in nujol mull or KBr on a Perkin-Elmer 842/Beckman IR-20 / Hitachi 215 spectrometers. The proton magnetic resonance spectra were recorded on a VXR-200 MHz or R-32 Perkin-Elmer 90 MHz spectrometer in CDC₁₃ or DMSO-d₆ using tetramethylsilane (TMS) as internal reference standard. The chemical shifts are expressed in δ (ppm) units downfield from TMS. Mass spectra were scanned on a Jeol JMX-DX-300 spectrometer operating at 70 eV. Carbon, hydrogen and nitrogen analyses were carried out on a Yanaco MT-3 (JAPAN) instrument. Thin layer chromatography (TLC) were performed on silica-gel plates using acetone-benzene (1 : 3 or 1 : 2) as solvent system and iodine chamber as visualizing agent.

Typical procedure for the synthesis of 6-Methoxyisatin-3-thiosemicarbazone(I)

A mixture of 6-methoxyisatin (0.18g, 0.001 mol) in Anhyd. ethanol (2ml) and thiosemicarbazide (0.1g, 0.0011 mol) in a mixture of water (2 ml) and glacial acetic acid (0.5 ml) was irradiated under microwave irradiation at 560W for 5-minutes. A yellow coloured solid formed during irradiation. The solid was filtered, washed well with water and crystallized from ethanol-DMF furnishing yellow crystals. yield 0.247g (95%), m.p. 265°C. [Found : N, 22.68, S,

12.62. $C_{10}H_{10}N_4O_2S$ requires N, 22.40; S, 12.80%; IR : 825, 860 (1, 2, 4-trisubstituted benzene ring), 1115 (C=S), 1125 & 1370 (C-O-C stretching), 1620 (C=N), 1700 (C=O), 3200, 3280, 3400 (NH, NH_2).

Typical procedure for the synthesis of 7-Methoxy-5H-2,3-dihydro [1, 2, 4] triazino [5, 6-b] indole-3-thione(II)

6-Methoxyisatin-3-thiosemicarbazone (I, 0.125g, 0.0005 mole) in 5% KOH (3.5 ml) was irradiated under microwave irradiation at 560W for 5-minutes. The reaction mixture was cooled and the insoluble material removed by filtration. The filtrate on neutralisation with dil. HCl gave a yellow solid which was filtered, washed well with water and crystallised from aq. DMF furnishing yellow crystals, yield 0.108 g (92%), m.p. > 260°C [Found : C, 51.91; H, 3.57; N, 23.84; S, 13.58. $C_{10}H_8N_4SO$ requires C, 51.72; H, 3.45; N, 24.13; S, 13.79%]; IR : 810, 860 (1, 2,

4-Trisubstituted benzene ring), 1150 (C=S), 1170, 1370 (C-O-C stretching), 1590, 1610 (C=N), 3200 (N-H stretching).

Typical procedure for the synthesis of 5H-3-(p-chlorophenacylthio-7-methoxy[1, 2, 4]triazino [5, 6-b] indole hydrobromide (IIIa; Ar = p-C₁C₆H₄)

A mixture of II(0.232 g, 0.001 mol) and p-chlorophenacyl bromide (0.234 g, 0.001 mol) in DMF (6 ml) was irradiated under microwave irradiation at 560W for 5-minutes, and poured into ice-water. The solid thus separated, was filtered, washed with water and crystallized from aq. DMF to give IIIa as yellow crystals, yield 0.415 g (94%), m.p. > 260 [Found: N, 12.20; S, 6.92. $C_{18}H_{14}N_4O_2SBrCl$ requires N, 12.03; S, 6.87%]; IR : 810, 860 (1, 2, 4-trisubstituted benzene ring), 1170, 1370 (C-O-C stretching), 1570 (C-N stretching), 1590, 1610 (C=NO, 1690 (C=O), 3180 (N-H stretching).

Table 1: 3-p-Chlorophenyl-8-methoxythiazolo [3', 2' : 2, 3][1, 2, 4] triazino [5, 6-b] indole

S. No.	Substrate(R)	Time(in minutes)	Yield(%)	m.p.(°C)
1.	-pClC ₆ H ₄ -(IIIa)	5	94	> 260°C
2.	-pBrC ₆ H ₄ -(IIIb)	5	91	> 260°C
3.	-C ₆ H ₅ (IIIc)	5	98	> 260°C

Table 2: 3-p-Chlorophenyl-8-methoxythiazolo [3', 2' : 2, 3][1, 2, 4] triazino [5, 6-b] indole

S. No.	Substrate(R)	Time(in minutes)	Yield(%)	m.p.(°C)
1.	-Cl(IVa)	5	91	> 250°C
2.	-Br(IVb)	5	90	> 250°C
3.	-H(IVc)	5	96	> 250°C

Following members of the series were also prepared in a similar way

IIIb (Ar = p-BrC₆H₄-), yield (91%), m.p. > 260°C [Found : N, 11.23; S, 6.38, $C_{18}H_{14}N_4O_2SBr_2$ requires, N, 10.98; S, 6.27%]; IR: 1570 (C-N stretching), 1610 (C=N), 1690 (C=O), 3190 (N-H). IIIc (Ar = C₆H₅), yield (98%) m.p. > 260°C [Found : N, 14.62; S, 8.18. $C_{18}H_{15}N_4O_2SCl$ requires N, 14.49; S, 8.27%]; IR : 1575 (C-N stretching), 1600 (C=N), 3240 (N-H stretching).

Typical procedure for the synthesis of 3-p-Chlorophenyl-8-methoxythiazolo[3', 2' : 2, 3][1, 2, 4] triazino [5, 6-b] indole(IVa, R = Cl)

Ketone IIIa(0.1g) in a mixture of H₃PO₄ (0.3ml) and P₂O₅ (0.4g) was irradiated under microwave irradiation at 560W for 5-minutes. The reaction mixture was poured into water and neutralised with aq. K₂CO₃ solution. The solid, thus separated, was filtered, washed well with water and crystallised from aq. DMF to furnish IVa as dark red crystals, yield 0.071 g (91%), m.p. > 250° [Found :

C, 59.17; H, 3.11; N, 15.16; S, 8.91. $C_{18}H_{11}N_4SOCl$ requires C, 58.93; H, 3.01; N, 15.27; S, 8.73%; IR; 1515 (C-N stretching) 1600 (C=N). PMR (DMSO- d_6) : 3.95 (3H, s, C_8-OCH_3), 7.40 [2H, d (J=7.5Hz), H-3' and H-5'], 7.70 [2H, d (J = 7.5.Hz), H-2' and H-6'], 7.90 (1H, s, C_2-H), 6.9-8.1 (3H, m, aromatic protons of indole moiety).

7.78%; IR : 1520 (C-N stretching), 1610 (C = N).

IVc (R=H) : yield (96%), m.p. > 250°C [Found : C, 64.92; H, 3.68; N, 16.64; S, 9.87. $C_{18}H_{11}N_4SO$ requires C, 65.06; H, 3.61; N, 16.86; S, 9.63%]; IR : 1520 (C-N stretching), 1610 (C = N).

Following members of the series were also prepared in a similar way

IVb (R = Br): yield (90%), m.p. > 250°C [Found : C, 52.38; H, 2.76; N, 13.42; S, 7.51. $C_{18}H_{11}N_4SOBr$ requires C, 52.55; H, 2.67; N, 13.62; S,

We thank Professor D. Villemin (France), Dr. R. Sharma (Dayton, USA) and Professor A.J. Bellamy (Swindon, UK) for inspiration.

ACKNOWLEDGMENTS

REFERENCES

- Jag Mohan, G.S.R. Anjaneyulu & Kiran, *Indian J. Chem.* **27B**: 128 (1988).
- H.R. Snyder, Jr. and L.E. Benjamin, *J. Med. Chem.*, **9**: 402 (1966).
- Norwich Pharmacal Co. *Neth. Appl.* **6**: 400,380 (1964); *Chem. Abstr.*, **62**: 2780(1965).
- D.L. Trepanier and P.E. Krieger, *U.S. Pat.*, **3**: 641,019 (1972); *Chem. Abstr.*, **76**: 127024k (1972).
- C.J. Sharpe, R.S. Shadbolt, A. Ashford and J.W. Ross, *J. Med. Chem.*, **14**: 977 (1971).
- M.M. KOchhar and B.B. Williams, *J. Med. Chem.*, **15**: 322 (1972).
- S. Kano and T. Noguchi, *Japan Pat.*, **71**: 37, 836 (1971); *Chem. Abstr.*, **76**: 25295g (1972).
- E.E. Renfrew and H.W. Pons, *U.S. Pat.*, **3**: 950,130 (1976); *Chem. Abstr.*, **85**: 22753e (1976).
- P.W. Jenkins and L.G.S. Brooker, *U.S. Pat.*, **3**: 681,081 (1972); *Chem. Abstr.*, **78**, 85949z (1973).
- M. Hinata, K. SHiba, H. Takei, A. Sato and T. Sakai, *Ger. Offen.*, **2**: 418, 278 (1974); *Chem. Abstr.*, **82**: 49815K (1975).
- L.G.S. Brooker, *U.S. Pat.*, **2**: 089,729 (1937); *Chem. Abstr.* **31**: 6989 (1937).
- (a) Romagnoli, R; Baraldi, P.G; Cruz-Lopez, C Preti, D Bermejo, J Estavez, *F. Chem. Med. Chem.* **4**: 1668 (2009).
(b) Bursavich, M. G Gilbert, A.M Lombardi, S Georgiadis, K. E. Reifenberg, E. Flannery, C. R; Morris, E.A. *bioorg. Med. Chem. Lett.* ,**17**: 5630 (2007).
(c) Konkel, M.J Packiarajan, M Chem. H Topiwala, U.P. Jimenez, H Talisman, I.J Coate, H Walker, M.W. *Bioorg. Med. Chem. Lett.* **16**: 3950 (2006).
(d) Lam, P.Y.S.Vinoent, G Clark, C.G Dcudon, S Jadhav, P.K. *Tetrahedron Lett.* **42**: 3415 (2001).
- (a) Shindikar, A.V Khan, F Viswanathan, *C.L. Eur.J.Med. Chem.* **41**: 786 (2006).
(b) Moser, P. Sallmann, A Wieserberg, *I.J. Med. Chem.* **33**: 2358 (1990).
(c) Sarges, R Howard, H. R Koe. H.K Weissman, *A.J.Med. Chem.* **32**: 437 (1989).
- Peet, N.J. *Heterocycl, Chem.* **17**: 1514 (1990).
- For reviews, see: (a) Chem, Y Larock, R.C. In *Modern Arylation Methods*, Ackerman, J. ED Wiley/ VCH New York, PP401 (2009).
(b) Sanz, R. *Org. Prep. Proced. Int.* **40**: 215 (2008).
- (a) Lin, Z; Larock, R. C. *J. Org. Chem.* , **71**: 3198 (2006).
(b) Lin, Z; Larock, R. C. *Org. Lett.* **5**: 4673 (2003).
(c) Lin, Z; Larock, R. C. *Org. Lett.* **6**: 99 (2004).
- Lin, Z; Larock, R. C. *J. AR.Chem. Sec.* **127**: 13112 (2005).

18. Pintori, D. G Greaney, *M.F.Org. Lett.* **12**: 168 (2010).
19. Yoshida, H Shirakawa, E₄ Honda, Y Hiyama, *T. Angew. Chem, Inted.* **41**: 3246 (2002).
20. Zhao, J Larock, R.C.*J.Org. Chem.*, **72**: 583 (2007).
21. Rogness D. C Larock, R.C. *Tetrahedron Lett*, **50**: 4003 (2009).
22. Lin, Z; Larock, R.C.J. *Org. Chem.* **71**: 3198 (2006).
23. M.A. Khan and S. Ahmad, *Orient J. Chem.*, **28**(2): 1061-1065 (2012).
24. S. Kumar, S. Yadav, S. Jadon, V. Kumar, A. M. Khedr and K.C. Gupta, *Orient J. Chem.*, **28**(4): 1833-1836 (2012).