



## Antimicrobial and Thermal Properties of Coating Systems Modified with ZnO Nanoparticle and its Hybrid Forms: (A Review)

TOLUTOPE O. SIYANBOLA<sup>1\*</sup>, ABIODUN F. AKINSOLA<sup>2</sup>, JOSEPH A. ADEKOYA<sup>1</sup>,  
OLAYINKA O. AJANI<sup>1</sup>, CYRIL O. EHI-EROMOSELE<sup>1</sup>, GRACE I. OLASEHINDE<sup>3</sup> and  
ADESOLA A. AJAYI<sup>3</sup>

<sup>1</sup>Department of Chemistry, Covenant University, P.M.B. 1023, Ota, Ogun State, Nigeria.

<sup>2</sup>Department of Chemistry, Ekiti State University, P.M.B. 5363, Ado-Ekiti, Nigeria.

<sup>3</sup>Department of Biological Sciences, Covenant University, P.M.B. 1023, Ota, Ogun State, Nigeria.

\*Corresponding Author E-mail: tolu.siyانبola@covenantuniversity.edu.ng

<http://dx.doi.org/10.13005/ojc/330102>

(Received: December 04, 2016; Accepted: February 08, 2017)

### ABSTRACT

This review examines the unparalleled chemical and physical properties of ZnO nanoparticles and its hybrid forms. The influence of these multifunctional materials within the polymeric matrix of organic coatings was discussed. The scanning electron microscope is seen to provide relevant information about the dispersion of the hybrid and composite coating systems. This review provides concise information about the antimicrobial and thermal stability of composites.

**Keywords:** Nanoparticles, hybrid, coatings, antimicrobial, thermal stability.

### INTRODUCTION

The concerns over the increase in worldwide epidemic and infectious diseases have accentuated the need to design antibiotics and promoting sanitary practices. World smallest life (microbes) are established to be the most virulent since the stone age. Diseases like malaria caused by protozoa (*Plasmodium spp*), pneumonia caused by bacterial, fungus and virus, White pox caused by *Serratia marcescens*, whooping cough (*Bordetella pertussis*) and tuberculosis (*Mycobacterium*

*tuberculosis*) top the list of world infectious diseases<sup>1</sup>. Microbes such as *Escherichia*, *Streptococci* and *Staphylococci* are responsible for most hospital (nosocomial) infections. The United States of American alone have a statistical record of two (2) million cases of hospital acquired infections and about 90,000 deaths per year<sup>2</sup>. The development of drugs such as antibiotics have helped to reduce the deadliness of the infections. Years after there applications (i.e. antibiotics), scientists have noticed the resistant strains of bacteria to designed antibiotics. The misuse, protracted use and bacterial

evolution was observed to have caused the resistant strain. The campaign against microbes have geared manufacturals towards formulating products like mouthwash, room spray, kitchen soaps, glass cleaner, shampoo, surface cleaners and perfume with antifungi and antibacterial chemicals<sup>3</sup>. The instinctual ability of bacteria to attach itself to surfaces and form thin colonies layers, which (for example) is capable of covering medical device surfaces and thereby creating medium of infection could be tackled by developing coating formulations able to inhibit its growth. Devices like catheters, mechanical heart valves, contact lenses, intrauterine devices, surgical consumables and orthopedic implants can become mediums for infection if not rightly protected<sup>4</sup>. There are various types of protection that could inhibit the growth of these microbes. This review primarily emphasis the formulation of eco-friendly coating systems modified with ZnO nanoparticles and its hybrid forms derived from seed oils. Coating systems are formulations uniquely designed for surface covering of substrates. They provides functional, decorative, and in some cases both on its applied substrate<sup>5</sup>. Plant seed oils based coating systems has recently been studied by researchers in a bid to substitute those derived from conventional petrochemicals base coating systems. Unlike petrochemical feed stocks that create environmental concerns ranging from air to water pollutions and global warming disaster; plant seed oils are renewable biodegradable resource materials that can be formulated into effective coating systems<sup>6-8</sup>. However, the nature of most plant seed oils (i.e. the composition of their fatty acid profiles) deters them from been used directly in coating formulations hence, the need to modify the seed oils in other to create reactive functional sites. Modified polyols systems prepared through aminolysis, transestrification, hydroformylation, epoxidation, partial glyceride (PG) formation, blown oil are pathways towards creating functional sites for coating systems<sup>9-11</sup>. Coating systems such as polyesteramides, polyetheramides, polyesteramide-urethanes are synthesized from plant base polyol platforms as exemplified in Scheme 1<sup>8, 10</sup>. The synthesized pristine organic coating products however shows limited thermal, corrosion, antimicrobial, rigidity, and chemical characteristic properties required in structural applications<sup>12-14</sup>. It is in the light of this, that researchers are incorporating designed nanoparticles within the

polymeric matrix of pristine coatings in order to reinforce the characteristic properties of the polymer composite thereby, creating innovative solutions in the area of optics, electronics, biomedical and material science. Quite a number of nanoparticles have been used in composite coating formulation processes. Commonly used based nanoparticles in this regard are TiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, CaCO<sub>3</sub>, ZnO, etc. The later (i.e. ZnO nanoparticle) possesses unique characteristics such as improved chemical, biological, semi-conductor properties and drug delivery. Nano-sized ZnO particles have also been found to be inert, non-toxic in nature and possess a blanket for UV radiation (due to its wide band gap of 3.37 eV, large bond strength and its heavy 60 meV excitation binding energy at room temperature)<sup>15-18</sup>. The inorganic amphoteric oxide (i.e. ZnO) is nearly insoluble in water and reacts slowly with fatty acids of triglycerides hence, the need to modify the periphery hydroxyls functional groups of the ZnO with organic compounds<sup>19-20</sup>. This will provide the required ambience for reaction between the triglyceride polyols, polyesters, polyethers and urethanes with the modified hybrid ZnO nanoparticles. This paper aims to review synthesis routes of preparing organic coating composites via the introduction of hybrid ZnO nanoparticles (ZnO-APTMS or ZnO-APTES) within the pristine polymer matrix and discussed thermal and corrosion stabilities, chemical resistance and antimicrobial sensitivity of seed oil based coating composites. Section 2 present basic discussions on ZnO nanoparticle and synthesis routes for preparation of hybrid-ZnO nanoparticles and its composites. In section 3, property evaluations (such as thermal stability, corrosion studies, chemical resistance and antimicrobial sensitivity) of coating composites were reviewed. The concluding remarks are presented in section 4.

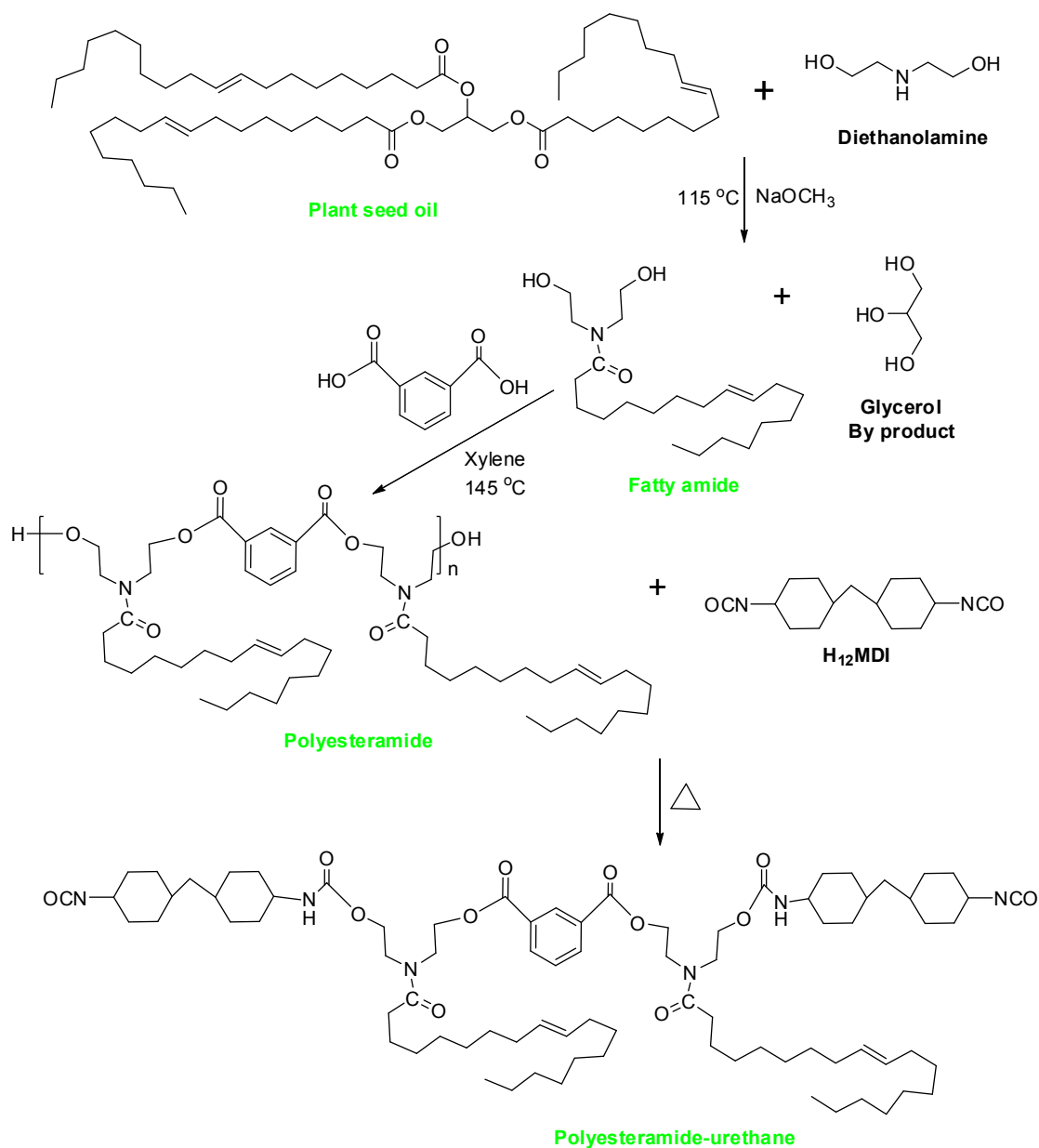
### ZnO nanoparticles

Over the years, ZnO nanoparticles have received extensive attention owing to its versatile industrial applications. The symmetric center deficiency in its wurtzite structure model (Figure 1) compounded with wide electromechanical coupling ensues in strong pyroelectric and piezoelectric properties. These effects results in the use of ZnO material in mechanical actuators and piezoelectric sensors<sup>21</sup>. This inorganic metal oxide (ZnO) with high thermal and mechanical stability at room

temperature belongs to group II-IV semiconductor and has its covalence between ionic and covalent semiconductors limits<sup>22-23</sup>.

The conductometric method of analysis shows that the antibacterial activity of CaO, MgO and ZnO can be ascribed to the reactive surface oxygen species on these oxides<sup>24</sup>. These inorganic oxides contain requisite mineral elements in human systems, which when administered in small amounts present

antimicrobial agent tendencies. This tendency is reported to be dependent on the particulate size of these oxides<sup>25</sup>. This multifunctional material (ZnO) creates new chemistries for organic base coatings. However, ZnO nanoparticles have significant tendency to agglomerate in coating systems. Hence, the need to modify the hydroxyl functional peripheral groups of ZnO nanoparticles especially via sol-gel method of creating hybrid materials (Scheme 2) for various industrial applications, such as the



preparation of adhesives and coatings<sup>26</sup>. This method is simple, reliable, low cost, and repeatable synthetic route that provides for its surface modifications<sup>27</sup>.

### Synthesis of hybrid-ZnO (ZnO-APTMS or ZnO-APTES) nanoparticles and its seed oil based composites

Recently, researchers have designed several methods of modifying the surface of ZnO material in a bid to reducing the effect of agglomeration and intercalation (when combined with other materials such as montmorillonite) without impairing the physicochemical properties of the compound<sup>28</sup>. The following sections present scientific

contributions towards modifying ZnO hybrid systems and composites formulations for coating purposes.

Dhoke *et al.*<sup>29</sup> reported an attempt to synthesize a nano-composite by incorporating nano-ZnO within a developed alkyd-based waterborne coating at different loading percentages. The composite were synthesized by combining alkyd-based waterborne coating with hexamethylolmelamine as cross-linker in the mixing ratio of 70:30. This base matrix coating was treated with 0.01, 0.02, and 0.03 wt.% nano-ZnO. Dispersion was carried out using mechanical stirrer initially then culminating with ultrasonication. The application of the coating was

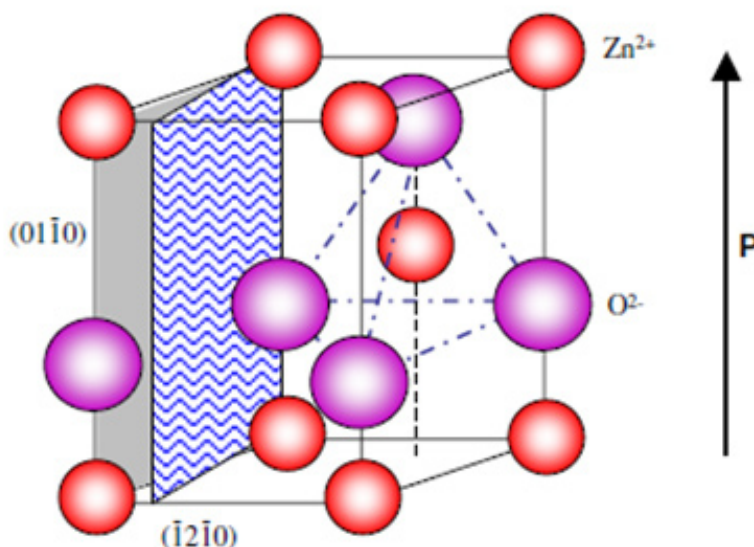
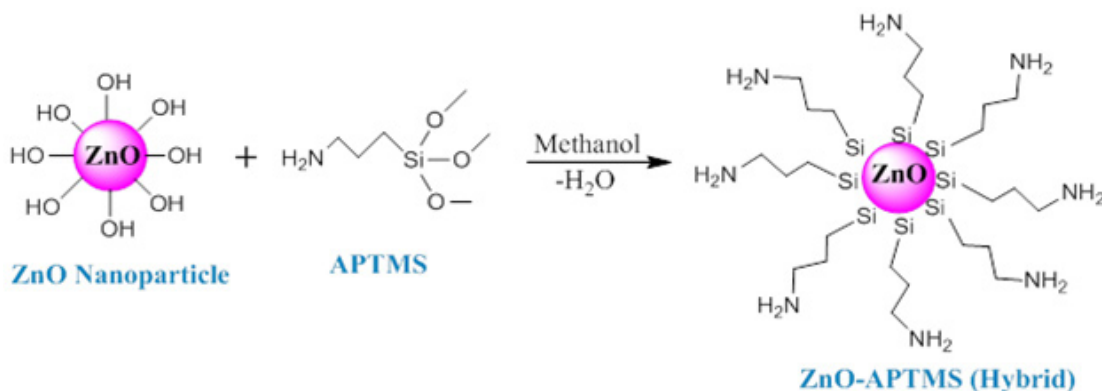


Fig. 1: Wurtzite structure model of ZnO <sup>17</sup>



Scheme 2: Surface modification of ZnO nanoparticles

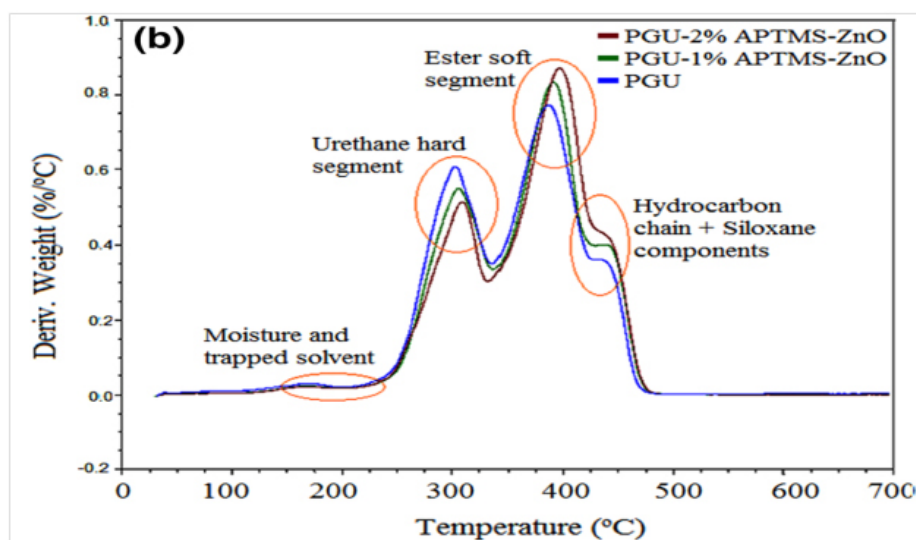
done via dipping of pretreated mild steel panels in the coating (nanocomposite). Curing was at 130 °C for 15 minutes. Similar procedure was reported by Dhoke *et al.*<sup>30</sup> were mechanical and heat-resistance properties of waterborne silicone-modified alkyd-based coatings were investigated alongside effect of nano-ZnO addition. Li *et al.*<sup>31</sup> prepared reinforced polyurethane coatings by mixing hydroxyl-acrylic resin (HAR) with an average molecular weight of 15,600 and hexamethylene-1,6-diisocyanate (HDI trimer). Planetary ball milling instrument was used in mixing along with percentages of ZnO nanoparticles. 30 g of thinner was mixed with some ZnO nanoparticles for 1.5 hour at 40 rpm. Subsequently, the suspension obtained was agitated with 40 g HAR and 10 g of HDI trimer. The mixture was later ball-milled for 1 hour at 40 rpm. Reinforced

polyurethane composite coatings were later prepared. Jena *et al.*<sup>32</sup> modified the peripheral of ZnO with 3-aminopropyl-triethoxysilane and afterwards incorporate the 3-aminopropyl- triethoxysilane-ZnO (APTES-ZnO) within the polymeric matrix of hyperbranched polyurethane-urea. This was achieved by adding 10 g of ZnO into a 50 g toluene in a round bottomed flask. The content was stirred with a magnetic stirrer for 5 minutes. Ultrasonic bath was used to improve the solubility of ZnO in toluene. 1 g of APTES coupling agent was added to the suspension and stirred at room temperature. After the appearance of a yellow-transparent dispersion the mixture was refluxed for 24 hours. Rotatory evaporator was used to dry the modified material. Unreacted APTES molecules were removed by washing with ethanol (3 times). The resultant power

**Table 1: Antimicrobial activity of pure polymer and PGU-APTMS-ZnO hybrid coatings<sup>33</sup>**

Sample	<i>Bacillus subtilis</i>	<i>Staphylococcus aureus</i>	<i>E. coli</i>	<i>Klebsiella pneumonia</i>	<i>Aspergillus niger</i>
PGU	-	-	+	-	-
PGU-1% APTMS-ZnO	-	+	+	++	N.D
PGU-1.5% APTMS-ZnO	-	++	++	+	N.D
PGU-2% APTMS-ZnO	-	+++	+++	++	+++

“+” indicates the sample showed antibacterial activity, “-” implies no antibacterial activity  
N.D indicates not determined



**Fig. 2: Differential thermogravimetric analysis (DTG) of PGU, PGU-1% APTMS-ZnO, PGU-2% APTMS-ZnO<sup>33</sup>**

was dried at 50 °C for 1 hour. Finally, the powder was ground and dried at 100 °C for 2 hours. In the same vein, Siyanbola *et al.*<sup>33</sup> also modified ZnO using 3-aminopropyl-trimethoxysilane.

### Characterizations

The plant seed oils, its polyols, urethanes and composites are usually structurally characterized by the following physico-chemical, spectroscopic and morphological examinations thus acid value, hydroxyl

value, iodine value, saponification value, refractive index, specific gravity, viscosity, gel permeation chromatography (GPC), Fourier Transform Infrared (FTIR), <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, scanning electron microscope (SEM), Transmission electron microscope (TEM), atomic-force microscope (AFM), X-ray diffraction analysis (XRD), thermogravimetric analysis (TGA), differential scanning calorimetry (DSC), dynamic mechanical thermal analysis (DMTA) etc.

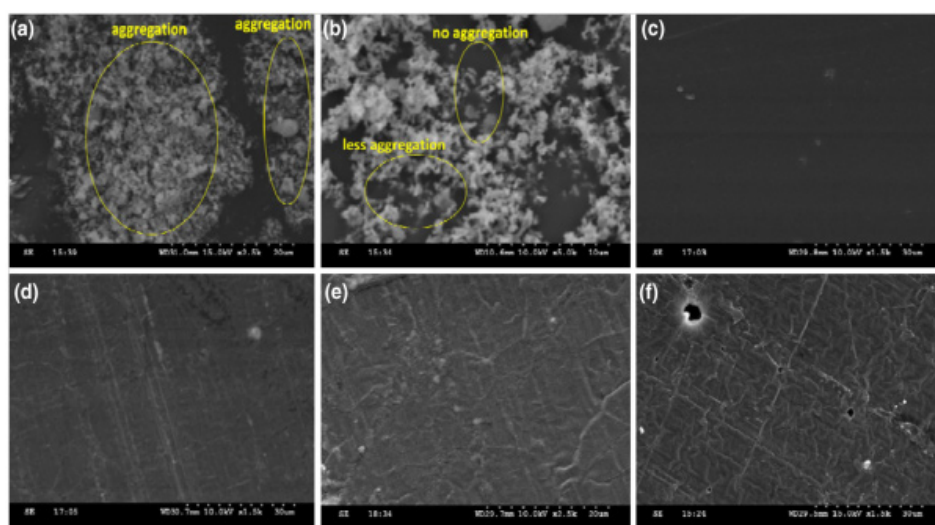


Fig. 3: SEM micrographs of ZnO (a), APTMS-ZnO (b), PGU (c), PGU-1 % APTMS-ZnO (d), PGU-1.5 % APTMS-ZnO (e), PGU-2 % APTMS-ZnO (f) hybrid films<sup>33</sup>

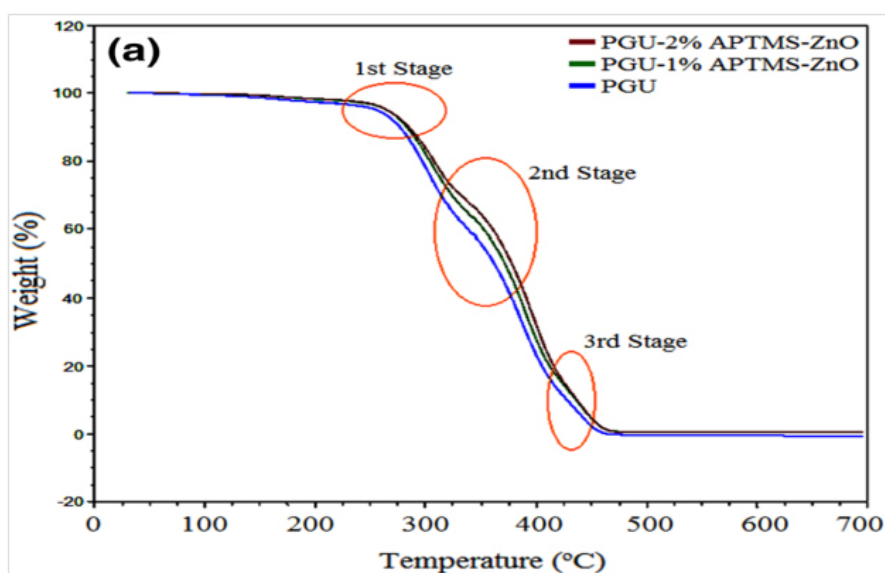


Fig. 4: Thermogravimetric Analysis (TGA)<sup>33</sup>

## Evaluations

### Antimicrobial sensitivity

Over the years, researchers have formulated different types of coatings systems from seed oil base feedstock, which have shown reasonable antimicrobial inhibitive tendency. Siyanbola *et al.*<sup>33</sup> compare the antimicrobial sensitivity of pristine polyurethane coating of a partial glyceride polyol and those that have their polymeric matrix doped with ZnO and modified ZnO (APTMS-ZnO) in varying percentages. The coating films were tested on gram positive (*Staphylococcus aureus* and *Bacillus subtilis*), gram negative (*Escherichia coli* and *Klebsiella pneumonia*) bacterial strains and a fungal stain (*Aspergillus niger*) grown on Czapek–Dox medium. The report shows inhibitive zone beneath and in the surrounding of the coating films. Polyurethane composite of 2% APTMS-ZnO was found to be highly resistive to growth on *Staphylococcus aureus* and *Escherichia coli*. *Bacillus subtilis* growth was not impaired by the placing of the coating in the medium (as shown on Table 1). The inhibitive activities of the hybrid coatings were perceived to be chiefly as a result of ZnO nanoparticle present in the composite films Chao *et al.*<sup>34</sup>. Similarly, Siyanbola *et al.*<sup>8</sup> carried out antibacterial activity on the same base seed oil plant (*Thevetia peruviana*) but on its fatty amide polyol [N,N'-bis (2-hydroxyethyl) *Thevetia peruviana* fatty amide] urethanes pristine and composites. Though, the synthesized composites in this case contain higher percentages of the hybrid materials, which also reflect on the degree of zone of inhibition. Furthermore, The antibacterial test carried out by Li *et al.*<sup>31</sup> on polyurethane resin composites vividly show the effect of ZnO nanoparticles percentage increase in the synthesized composites. As the percentage weight of ZnO increases in the coating system the zone of inhibition increases especially in *E. coli*.

### Thermal stability

The differential thermogravimetric analysis (DTG) reveals that synthesized coating composites with ZnO and its modified forms usually have a

two stage degradation steps<sup>8,35</sup>. The contrary, were recorded with systems developed by Siyanbola *et al.*<sup>33</sup> with a three stage degradation steps (shown in Figure 2). This observation may be due to surface morphology achieved during the coating formulation (Figure 3), which may also be in regard to the material percentage composition within the polymer matrix. The thermal stability of ZnO and its hybrid coating composites generally increases as the percent composition of materials increases (Figure 4)<sup>33,35</sup>. Siyanbola *et al.*<sup>33</sup> reported 278.85 °C onset degradation step for PGU-2% APTMS-ZnO while PGU-1 % APTMS-ZnO shows 22.70 °C less than PGU-2% APTMS-ZnO. The storage modulus of these coating films also reflects the influence of hybrid composites in the pristine polyurethane. The modulus profile formation is as a result of strong network structures between the urea groups and surface hydroxyl groups on the polyurethane composites through hydrogen bonding. Mishra *et al.*<sup>35</sup> also synthesized aqueous polyurethane hybrid dispersion using nano ZnO as filler. The research report corroborates the findings about higher thermal stability for hybrid films than that of mother (pristine) polymer.

## CONCLUSION

The peripheral hydroxyl functional group surrounding ZnO confers on this oxide robust modification pathways. These product pathways developments have led to the formulation of different coating composites (when the hybrid material is dispersed within the pristine polymeric coatings). These composites are capable of impeding antimicrobial growth and stabilizing thermal interferences.

## ACKNOWLEDGEMENTS

The author Dr. Tolupe O. Siyanbola appreciates Covenant University, Ota, Ogun State, Nigeria for creating true research platform for her faculties.

## REFERENCES

1. World Health Statistics 2009, *World Health Organization*, **2009**.
2. Gabriel, G. J.; Som, A.; Madkour, A. E., Eren, T.; Tew, G. N.; *Mater. Sci. Eng. R: Rep.* **R2007**, *57*, 28-64.
3. Alex, K.; Shane S.; Bret J. C.; *Prog. Org. Coat.*,

- 2011, 72, 222-253.
4. Donlan, R. M.; Biofilms and Device-Associated Infections. *Emerging Infectious Diseases*. **2001**, 7(2), 277-281.
  5. Fristad, W. E.; "Epoxy Coatings for Automotive Corrosion Protection". **2000**, doi:10.4271/2000-01-0617
  6. Lu, Y.; Larock, R. C.; *ChemSusChem.*, **2009**, 2, 136-147.
  7. Siyanbola, T. O.; Ajanaku, K. O.; James, O. O.; Olugbuyiro, J. A. O.; Adekoya, J. A.; *Glob J. Pure Appl. Sci. Tech.*, **2011**, 1, 49-54.
  8. Siyanbola, T. O.; Sasidhar, K.; Anjaneyulu, B.; Kumar, K. P.; Rao, B. V. S. K.; Narayan, R.; Olaofe, O.; Akintayo, E. T.; Raju, K. V. S. N.; *J. Mater. Sci.*, **2013**, 48, 8215-8227.
  9. Özgür, Seydibeyođlu M.; Misra, M.; Mohanty, A.; Blaker, J. J.; Bismarck, A.; Kazemizadeh, M. J.; *Mater. Sci.*, **2013**, 48, doi 10.1007/s10853-012-6992-z
  10. Meier, M. A. R.; Metzgerb J.O. and Schubert U.S., Plant oil renewable resources as green alternatives in polymer science. *Chem. Soc. Rev.*, **2007**, 36, 1788-1802.
  11. Sharmin, E.; Ashraf, S.M.; Ahmad, S.; *Eur. J. Lipid Sci. Technol.*, **2007**, 109, 134-146.
  12. Uyama, H.; Kuwabara, M.; Tsujimoto, T.; Nakano, M.; Usuki, A.; Kobayashi, S.; *Macromol. Biosci.*, **2004**, 4, 354-360.
  13. Changqing, Fu; Zhe, Yang; Zitong, Zheng; Liang, Shen; *Prog. Org. Coat.*, **2014**, 77, 1241-1248.
  14. Jayakumar, R.; Rajkumar, M.; Nagendran, R.; Nanjundan, S.; *J. Appl. Polym. Sci.*, **2002**, 85, 1194.
  15. Fangli, Y.; Peng, H.; Chunlei, S. Y.; Shulan, S. H.; Jinlin, L.; *J. Mater. Chem.*, **2003**, 13, 634-637.
  16. Nasuaad, A.; Otsego, Y.; *J. Colloid Interface Sci.*, **2006**, 296(2), 558-564.
  17. Dhoke, S. K.; Khanna, A. S.; Mangal, Sinha T. Jai; *Prog. Org. Coat.*, **2009**, 64, 371-382.
  18. Li-Piin, Sung; Stephanie, Scierka; Mana, Baghai-Anaraki; Derek, L. H.; Characterization of Metal-Oxide Nanoparticles: Synthesis and Dispersion in Polymeric Coatings *Mat. Res. Soc. Symp. Proc.* **2013**. 740, 15.4.0-15.4.6
  19. Hernandezbattez, A; Gonzalez, R; Viesca, J; Fernandez, J; Diazfernandez, J; MacHado, A; Chou, R; Riba, J.; *Wear*, **2008**, 265(3-4), 422. doi:10.1016/j.wear.2007.11.013
  20. Greenwood Norman N.; Earnshaw, *Alan Chemistry of the Elements* (2nd ed.). Butterworth-Heinemann. **1997**, ISBN 0-08-037941-9.
  21. Zhong, Lin Wang; *J. Phys. Condens. Matter.*, **2004**, 16, R829-R858.
  22. Ko<sup>3</sup>odziejczak-Radzimska; Teofil Jesionowski.; *Materials*, **2014**, 7, 2833-2881; doi:10.3390/ma7042833.
  23. Wang, J.; Cao, J.; Fang, B.; Lu, P.; Deng, S.; Wang, H.; *Mater. Lett.* **2005**, 59, 1405-1408.
  24. Sawai, J.; Yoshikawa, T. J.; *Appl. Microbiol.*, **2004**, 96, 803
  25. Padmavathy N.; Vijayaraghavan, R.; *Sci. Technol. Adv. Mater.*, **2008**, 9, 1-7.
  26. Jena, K. K.; Narayan, R.; Raju, K. V. S. N.; *Polym. International*, **2012**, 61(8), 1309-1317.
  27. Mahato, T. H.; Prasad, G. K.; Acharya, B. S. J.; Srivastava, A. R.; Vijayaraghavan, R.; *J. Hazard. Mater.*, **2009**, 165, 928-932.
  28. Mohammed H.; Abdul Latif; *Chemistry and Materials Research*. **2015**, 7(5), 18-23.
  29. Dhoke, S. K.; Khanna, A. S.; Sinha, M. T.; *Prog. Org. Coat.*, **2009**, 64, 371-382.
  30. Dhoke, S. K.; Bhandari R.; Khanna, A. S.; *Prog. Org. Coat.*, **2009**, 64, 39-46.
  31. Li, J. H.; Hong, R. Y.; Li, M. Y.; Li, H. Z.; Zheng, Y.; Ding, J.; *Prog. Org. Coat.*, **2009**, 64, 504-509.
  32. Jena, K. K.; Rout, T. K.; Narayan, R.; Raju, K. V. S. N.; *Polym. Int.*, **2012**, 61: 1101-1106.
  33. Siyanbola, T. O.; Sasidhar, K.; Rao, B. V. S. K.; Narayan, R.; Olaofe, O.; Akintayo, E. T.; Raju, K. V. S. N.; *J. Amer. Oil Chem. Soc.*, **2015**, 92, 267-275.
  34. Chao, W.; Lian-Long, L.; Ai-Ting, Z.; Peng, X.; Jian-Jun, L.; Xiao-Ting. Z.; *Afri. J. Biotech.*, **2012**, 11, 10248-10254.
  35. Mishra, A. K.; Mishra, R. S.; Narayan, R., Raju, K. V. S. N.; *Prog. Org. Coat.*, **2010**, 67, 405-413.