



A New Method for Synthesis of 2,4,5-Triaryl-1*H*-Imidazole Derivatives using SiO₂-NaHSO₄ under Solvent-free Conditions

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<http://dx.doi.org/10.13005/ojc/300143>

(Received: December 13, 2013; Accepted: January 10, 2014)

ABSTRACT

A mixture of benzyl, aromatic aldehyde and ammonium acetate in presence of SiO₂-NaHSO₄ under solvent-free condition were converted to 2,4,5-Triaryl-1*H*-Imidazoles. The short reaction time, cleaner reaction, and easy workup make this protocol practical and economically attractive.

Key words: Imidazoles, Solvent-free, Multicomponent reactions, SiO₂-NaHSO₄

INTRODUCTION

Multicomponent reactions (MCRs) refers to a reaction in which two or more ingredients are combined within a single process and the products they create, which is part of all the components are present¹⁻³. Since the multi-component reactions for the synthesis of organic compounds and these compounds can be used as a drug and precursor multicomponent reactions, so to investigate them out is important⁴.

Imidazoles are important heterocycle compounds in medicinal chemistry. Imidazoles widely have been used of biological activity

which has made them privileged structures in combinational drug discovery libraries. They have the biological roles and also found application as a chromophore with high extinction coefficient, readily tunable absorption wavelength, and fluorophoric properties and was desirable as a large planar synthetic building block in supramolecular chemistry. Recently, several improved methodologies have been developed that use HY/silica gel, acidic Al₂O₃, AcOH, ionic liquid, NH₄OAc, sodium bisulfate among others⁵⁻⁸. Previously, we have synthesized a number of heterocyclic compounds⁹⁻¹⁷.

In this study, we have used of SiO₂-NaHSO₄ as a catalysts to develop a new and easy methodology

for the synthesis of 2,4,5-triaryl imidazole derivatives. The experiments were started with the study of one-pot, a simple, mild and efficient method for the preparation of the 2,4,5-triaryl imidazoles by using $\text{SiO}_2\text{-NaHSO}_4$ as a catalyst (Scheme1).

EXPERIMENTAL

All chemicals were obtained from Merck or Fluka without further purification. Silica gel SILG/UV 254 plates were used for TLC. IR spectra were measured on a Shimadzu IR-470 Spectrophotometer. ^1H NMR spectra were determined on Bruker 500 DRX AVANCE instrument at 500 MHz, respectively.

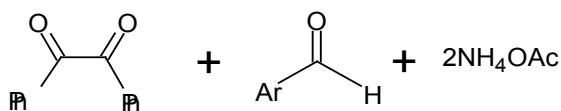
General procedure for preparation of 2a-I

A mixture of aldehyde (1 mmol), benzil (1 mmol), ammonium acetate (3 mmol) and $\text{SiO}_2\text{-NaHSO}_4$ (10 mol %) as a catalyst was stirred at 120 °C for 30 min. The progress of reaction was monitored by TLC. After finishing, recrystallized from ethanol 95% to give pure products (A1-A4)

spectral data

2,4,5-Triphenyl-1H-imidazole (A1)

Pale yellow crystals, Yield: 0.26 g (89%), mp: 274-276°C. IR ($\lambda_{\text{max}}/\text{cm}^{-1}$)(KBr): 3400(NH Str.);



^1H NMR (400.13 MHz CDCl_3) δ (ppm): 7.39-7.85(10H, m, CH_{arom}); 7.98-8.0(2H, d, $J=8\text{Hz}$, 2CH); 8.10(1H, s, CH).

RESULTS AND DISCUSSION

We have been able to introduce an efficient and environmentally friendly for the synthesis of imidazole derivatives via condensation of benzil with various aromatic aldehydes and ammonium acetate.

3000(CH_{arom} Str.); 1600($\text{C}=\text{C}$ Str.); 1470($\text{C}=\text{N}$ Str.). ^1H NMR (400.13 MHz CDCl_3) δ (ppm): 7.22-7.49(14H, m, CH_{arom}), 7.94(2H, d, $J=8\text{Hz}$, 2CH), 9.40(H, s, NH).

2-(4-Chlorophenyl)-4,5-diphenyl-1H-imidazole (A2)

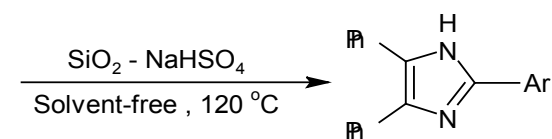
Bright crystals, Yield: 0.28 g (85%), mp: 260-262°C. IR ($\lambda_{\text{max}}/\text{cm}^{-1}$)(KBr): 3400(NH Str.); 3000(CH_{arom} Str.); 1600($\text{C}=\text{C}$ Str.); 1500($\text{C}=\text{N}$ Str.). ^1H NMR (400.13 MHz CDCl_3) δ (ppm): 7.09-7.84 (11H, m, CH_{arom}), 7.82-7.84(2H, d, $J=8\text{Hz}$, 2CH), 9.53(H, s, NH).

2-(4-Methoxyphenyl)-4,5-diphenyl-1H-imidazole (A3)

Pale yellow crystals, Yield: 0.28 g (86%), mp: 227-230°C. IR ($\lambda_{\text{max}}/\text{cm}^{-1}$)(KBr): 3442(NH Str.); 3070(CH_{arom} Str.); 1599($\text{C}=\text{C}$ Str.); 1450($\text{C}=\text{N}$ Str.). ^1H NMR (400.13 MHz CDCl_3) δ (ppm): 3.84(3H, s, CH_3); 6.93(2H, d, $J=8\text{Hz}$, 2CH); 7.27-7.53 (10H, m, 10CH), 7.82(2H, d, $J=8\text{Hz}$, 2CH).

2-(3-Nitrophenyl)-4,5-diphenyl-1H-imidazole (A4)

Pale yellow crystals, Yield: 0.3 g (88%), mp > 300°C. IR ($\lambda_{\text{max}}/\text{cm}^{-1}$)(KBr): 3400(NH); 3055 (CH_{arom}); 1600($\text{C}=\text{C}$); 1475($\text{C}=\text{N}$); 1420, 1375(NO_2).



Therefore, reported $\text{SiO}_2\text{-NaHSO}_4$ as catalyst which could provide an efficient, cheap, easy separation, high yield and simple route under solvent-free condition for the synthesis of imidazoles.

ACKNOWLEDGEMENTS

We gratefully acknowledge the financial support from the Research Council of Tonekabon Branch Islamic Azad University.

REFERENCES

1. Domling A., *Chem. Rev.*, **106**: 17 (2006).
2. Khan A. J and Basheer M., *Orient. J.Chem.*, **27**(4): 1759-1762 (2011).
3. Setamdideh D., Karimi Z and Rahimi F., *Orient. J.Chem.*, **27**(4): 1621-1634 (2011).
4. Kalinski C., Lemoine H., Schmidt J., Burdack C., Kolb J., Umkehrer M. and Ross G., *Syn. lett.*, **24**: 4007 (2008).
5. Gadekar L. S., Mane S. R., Katkar S. S., Arbad B. R. and Lande M. K., *Cent. Eur. J. Chem.*, **7**: 550 (2009).
6. Radziszewski B., *Chem. Ber.*, **15**: 1493 (1882).
7. Janvier P., Sun X. and Bienayme H., *J. Am. Chem. Soc.*, **124**: 2560 (2002).
8. Sridhara M. B., Srinivasa G. R. and Gowada D. C., *J. Chem. Res. Soc.*, **1**: 74 (2004).
9. Azizian J., Hatamjafari F., Karimi A. R. and Shaabanzadeh M., *Synthesis*. **5**: 765 (2006).
10. Azizian J., Shaabanzadeh M., Hatamjafari F. and Mohammadizadeh M.R., *Arkivoc.* (xi): **47** (2006).
11. Hatamjafari F., *Synthetic Communications*. **36**: 3563 (2006).
12. Azizian J., Hatamjafari F. and Karimi A. R., *Journal of Heterocyclic Chemistry*. **43**: 1349 (2006).
13. Hatamjafari F and Montazeri N., *Turkish Journal of Chemistry*. **33**: 797 (2009).
14. Hatamjafari F., *Orient. J. Chem.*, **28**: 141 (2012).
15. Hatamjafari F., *Orient. J. Chem.*, **29**: 93 (2013).
16. Hatamjafari F and Alijanichakoli F., *Orient. J.Chem.*, **29**: 145 (2013).
17. Hatamjafari F and Hosseinian A., *Orient. J.Chem.*, **29**: 109(2013).